April 7, 2014

Renee Robinson, P.G.
NuStar Energy, L.P.
19003 IH-10 West
San Antonio, Texas 78257

RE: SVE Evaluation and Proposed Soil Investigation Report
NuStar Pipeline Release North Andover Site
Andover, Kansas

Dear Ms. Robinson:

The Kansas Department of Health and Environment (KDHE) acknowledges receipt of the above-referenced document, prepared by Apex Companies, LLC, on behalf of NuStar Pipeline Operating Partnership, L.P. (NuStar), dated March 28, 2014. As indicated in the letter dated April 3, 2014, KDHE completed review of the Proposed Soil Investigation portion of the document and approved the work plan, indicating comments would be provided under separate cover to address the Soil Vapor Extraction (SVE) Evaluation.

Given the community’s resistance to SVE, KDHE has decided not to review the SVE portion of the report at this time. If the technology is considered a viable option in the future, KDHE will then review it. The document is admitted into the administrative file record for informational purposes.

No response to this letter is necessary. Should you have any questions regarding this letter, please contact me at 785-296-1935 or pgreen@kdheks.gov.

Sincerely,

Pamela Green
Environmental Scientist
Site Restoration Unit
Bureau of Environmental Remediation

c: Chris Carey → file C2-008-72928
    Grant Harse, KDHE Office of Legal Services
    Sam Jackson, Apex Companies, LLC
April 3, 2014

Renee Robinson, P.G.
NuStar Energy, L.P.
19003 IH-10 West
San Antonio, Texas 78257.

RE: SVE Evaluation and Proposed Soil Investigation Report
NuStar Pipeline Release North Andover Site
Andover, Kansas

Dear Ms. Robinson:

The Kansas Department of Health and Environment (KDHE) acknowledges receipt of the above-referenced document, prepared by Apex Companies, LLC, on behalf of NuStar Pipeline Operating Partnership, L.P. (NuStar), dated March 28, 2014. KDHE has completed review of the Proposed Soil Investigation portion of the document and approves the work plan. KDHE will provide comments under separate cover to address the SVE Evaluation.

No response to this letter is necessary. Please proceed with implementation of the proposed soil investigation as soon as possible. Should you have any questions regarding this letter, please contact me at 785-296-1935 or pgreen@kdheks.gov.

Sincerely,

Pamela Green
Environmental Scientist
Site Restoration Unit
Bureau of Environmental Remediation

c: Chris Carey → file C2-008-72928
Grant Harse, KDHE Office of Legal Services
Sam Jackson, Apex Companies, LLC
March 28, 2014

Ms. Pamela Green
Kansas Department of Health and Environment
Bureau of Environmental Remediation
1000 SW Jackson Street, Suite 410
Topeka, Kansas  66612

Re:  SVE Evaluation and Proposed Soil Investigation Report
    NuStar Andover Quail Crossing
    Andover, Kansas
    1641-04

Dear Ms. Green:

Enclosed, please find the SVE Evaluation and Proposed Soil Investigation Report (Report). This Report was prepared by Apex Companies, LLC (Apex) on behalf of NuStar Pipeline Operating Partnership L.P. (NuStar) in response to a gasoline release from a NuStar refined petroleum pipeline. As discussed earlier this week, NuStar is prepared to implement the proposed soil investigation as soon as April 7, 2014.

If you have any questions or would like to discuss this further, please contact me at (503) 924-4704 ext. 111

Sincerely,

Sam Jackson
Associate Engineer

ATTACHMENT
SVE Evaluation and Proposed Soil Investigation Report

cc:  Ms. Renee Robinson, NuStar Energy, L.P. (electronic deliverable)
SVE Evaluation and Proposed Soil Investigation Report
Quail Crossing Neighborhood
Andover, Kansas

Prepared for:
NuStar Pipeline Operating Partnership L.P.

March 28, 2014
1641-04

Sam Jackson
Associate Engineer

Chris Breemer, P.G.
Principal Geologist
Table of Contents

1.0 INTRODUCTION .............................................................................................................................. 1
  1.1 Objectives ..................................................................................................................................... 2
  1.2 Report Organization ........................................................................................................................ 2

2.0 BACKGROUND .................................................................................................................................. 2
  2.1 Site Description ................................................................................................................................ 3
  2.2 IRM and CI Activities ...................................................................................................................... 3

3.0 SVE PILOT TEST RESULTS ............................................................................................................. 4
  3.1 Pilot Test and Evaluation .................................................................................................................. 4
  3.2 System Layout and Specifications .................................................................................................... 5
  3.3 System Operation ............................................................................................................................. 7
  3.4 Schedule ......................................................................................................................................... 8

4.0 NEIGHBORHOOD MEETING AND PUBLIC OUTREACH .................................................................. 8

5.0 SOIL INVESTIGATION ....................................................................................................................... 9
  5.1 Proposed Soil Boring Locations ..................................................................................................... 10
  5.2 Preparatory Activities .................................................................................................................... 10
  5.3 Procedures ..................................................................................................................................... 10
  5.4 Drilling and Soil Sample Collection ............................................................................................... 10
  5.5 Soil Sampling and Analysis ............................................................................................................. 10
  5.6 Field Screening ............................................................................................................................... 11
  5.7 Quality Assurance and Quality Control ......................................................................................... 11
  5.8 Investigation-Derived Waste .......................................................................................................... 11

6.0 SCHEDULE ....................................................................................................................................... 11

7.0 REFERENCES ................................................................................................................................. 12

Table

1. SVE Pilot Test Effluent Data

Figures

1. Site Location Map
2. Site Plan and Sampling Location Map
3. Soil Vapor Extraction Test Well Locations
4. Conceptual Soil Vapor Extraction System
5. Soil Vapor Extraction System Schematic
6. Conceptual SVE Enclosure
7. Proposed Soil Boring Locations
Appendices

A  Field Notes and SVE Pilot Test Flow Rate Data
B  Analytical Laboratory Report – SVE Effluent Data
C  Sampling and Analysis Plan
1.0 Introduction

This Soil Vapor Extraction (SVE) Evaluation and Proposed Soil Investigation Report was prepared by Apex Companies, LLC (Apex) on behalf of NuStar Pipeline Operating Partnership L.P. (NuStar) in response to a gasoline release from a NuStar refined petroleum pipeline. The pipeline release was discovered after separate-phase hydrocarbons (SPH) were detected in an irrigation well at 2006 N Colt Court, in the Quail Crossing Neighborhood (the Neighborhood) of Andover, Kansas (the Site; Figure 1). The NuStar pipeline release location is approximately 80 feet north of the affected irrigation well.

Upon notification of the discovery of SPH in the irrigation well, NuStar immediately mobilized to the Site and implemented several initial response and abatement activities. Following the initial response activities, NuStar: (1) performed a number of interim remedial measures (IRMs); and (2) planned and implemented a Comprehensive Investigation (CI) in coordination with Kansas Department of Health and Environment (KDHE) and the Quail Crossing Neighborhood Association. These activities were performed in accordance with the Consent Agreement and Final Order (CAFO), dated May 10, 2013, and are described in detail in the CI Report (Apex, 2014). A groundwater monitoring program and weekly vacuum extraction events are ongoing at the Site.

Although a significant amount of soil containing gasoline constituents was removed during pipeline repair activities, information developed during the CI indicates that residual gasoline constituents remain in soil at some locations in the vicinity of the pipeline release. This material may be an ongoing source of gasoline constituents to groundwater. To evaluate the feasibility of SVE technology for removing gasoline constituents at the pipeline release area, NuStar performed an SVE pilot test during the week of December 9, 2013. This pilot test was performed in accordance with the KDHE-approved SVE Pilot Test Work Plan (Apex, 2013c). The methods and results of the pilot test, and a complete evaluation of the pilot test results, are presented herein.

A preliminary evaluation of the SVE pilot test results indicated that SVE may be an effective remedial technology to remove source area material in the vicinity of the release location. Accordingly, NuStar presented the preliminary results of the SVE pilot test and a conceptual SVE system design at a KDHE-organized Neighborhood meeting on February 17, 2014. In the week after the meeting, the Neighborhood evaluated the proposal and then stated that they would not approve of the construction and operation of an SVE system in the Neighborhood. Neighborhood representatives cited a number of reasons for not accepting an SVE system, including concerns about noise, emissions, and the appearance of above-ground infrastructure.

In an effort to evaluate other remedial alternatives that may be acceptable to Neighborhood residents and effective for removing hydrocarbon mass from the subsurface, NuStar proposes additional soil investigation in the vicinity of the pipeline release. The purpose of the additional soil investigation is to obtain data that
will be useful for refining the understanding of the extent of gasoline constituents in soil. These data will be used to select and design a remedial alternative that is acceptable to the Neighborhood.

1.1 Objectives

The objectives of this report are: (1) document the results of the SVE pilot test that was performed in December 2013; (2) present a conceptual SVE system design; (3) document outreach activities to the Neighborhood and the Neighborhood response to the proposed SVE system; and (4) propose additional activities to refine the understanding of the nature and extent of gasoline constituents in soil at the pipeline release area.

1.2 Report Organization

The following sections of this report include:

- **Background and Site Description** – This section describes the Site characteristics and findings of the CI and describes the interim remedial measures (IRMs) that have been implemented at the Site.

- **Soil Vapor Extraction Test Results** – This section presents the methods and results of the SVE pilot test and a design for a conceptual SVE system at the Site.

- **Neighborhood Feedback** – This section describes NuStar and KDHE outreach efforts to the Neighborhood regarding the potential installation of an SVE system, and the Neighborhood response to the proposed SVE system.

- **Proposed Activities** – This section describes investigation activities that are proposed to further refine the understanding of the nature and extent of gasoline constituents in soil; these data will be used, in coordination with KDHE, to evaluate alternative remedial options that may be acceptable to the Neighborhood.

2.0 Background

The gasoline release from the NuStar pipeline was discovered on June 8, 2012, following the detection of SPH in an irrigation well at the property at 2006 N Colt Court. Upon notification of the discovery of SPH in the irrigation well, NuStar immediately mobilized to the Site and implemented several initial response and abatement activities, including: (1) inspecting and testing of the pipeline; (2) hydrotesting of the affected portion of the pipeline; (3) excavation and removal of soil containing gasoline constituents along a 45-foot section of pipeline; (4) replacement of a portion of the pipeline; (5) collection and analysis of water samples from the irrigation well at 2006 N Colt Court; (6) field screening of nearby irrigation wells; (7) vacuum and manual removal of SPH and water from the irrigation well at 2006 N Colt Court; and (8) deactivation of the irrigation well at 2006 N Colt Court and irrigation wells at other properties, and connection of the associated...
irrigation systems to the municipal water supply. Many of these activities were discussed in the Initial Response Summary Report (Apex, 2012a) as well as in the Comprehensive Investigation Report (Apex, 2014).

Following the initial response activities, NuStar planned and implemented the CI and performed a number of IRMs in coordination with KDHE and the Quail Crossing Neighborhood Association. These activities are described in detail below.

2.1 Site Description

The Site is located in the northern portion of the City of Andover, in southwest Butler County, Kansas (Figure 1). Land use at the Site and surrounding areas is residential. As shown on Figure 2, a stormwater retention pond is located near the Site, approximately 250 feet south of the NuStar pipeline.

The NuStar pipeline is constructed of 8-inch-diameter carbon steel and is used to transport liquefied petroleum products. The pipeline was installed to an approximate depth of 3 feet below the ground surface (bgs) in agricultural land in 1959. The pipeline remained in agricultural and/or undeveloped land until the late 1990s when development of the Quail Crossing Neighborhood began.

As part of the Neighborhood development, the pipeline depth was increased to approximately 6 feet bgs by re-excavating and backfilling with native material. An approximately 50-foot-wide pipeline right of way (ROW) was created along the pipeline pathway for safety purposes during Neighborhood development. The ROW limits development activities near the pipeline. The pipeline location and associated ROW are shown on Figure 2.

The Neighborhood is located on 21st Street between 159th Street and Andover Road in the City of Andover, and encompasses approximately a one-quarter- by one-half-mile area. The Neighborhood is currently zoned for residential use (R-2) by Butler County. Approximately 165 homes are present in the Neighborhood. Utilities in the Neighborhood (power, natural gas, water, and sewer) are underground. The general location of underground utilities in the vicinity of the pipeline release is shown on Figure 2. Utilities will be mapped in greater detail during future work at the Site.

2.2 Interim Remedial Measures and Comprehensive Investigation Activities

In addition to the initial response and abatement activities discussed above, ongoing IRMs include weekly vacuum extraction of SPH and water from the irrigation well at 2006 N Colt Court. Since removal measures were initiated in June 2012, SPH levels have decreased significantly; and since August 2013, SPH have been only intermittently present in the irrigation well. SPH have not been detected in other wells in the Neighborhood.
The CI was performed following the initial response activities. Activities completed during the CI included:
(1) the advancement of 24 soil borings; (2) collection and analysis of soil samples from 16 soil borings; (3) installation of 16 groundwater monitoring wells, three soil vapor monitoring points, and four soil vacuum extraction (SVE) test wells; (4) collection and analysis of water samples from the 16 monitoring wells and 13 irrigation wells during one monitoring event and 16 irrigation wells during a second monitoring event; and (5) collection and analysis of vapor samples from the soil vapor monitoring points. The results of the CI were presented in the Comprehensive Investigation Report submitted in February 2014. During the CI, NuStar also performed a pilot test to evaluate if SVE would be an effective remediation technology for the Site.

3.0 SVE Pilot Test Results

To evaluate the feasibility of SVE technology for remediating gasoline constituents in the subsurface, NuStar performed an SVE pilot test on December 9 and 10, 2013. This pilot test was performed in accordance with the KDHE-approved SVE Pilot Test Work Plan (Apex, 2013c). The methods and results of the pilot test are presented below.

3.1 Pilot Test and Evaluation

The SVE pilot test was performed using pilot test wells that were installed specifically for pilot testing purposes (i.e., wells SVE-2, SVE-3, and SVE-4), and using appropriately constructed groundwater monitoring wells (that is, wells with adequate sections of the screen interval exposed to vadose zone soil). Details regarding the construction of these wells were presented in the CI Report (Apex, 2014).

The SVE pilot test consisted of two primary elements: (1) vacuum was induced individually at three SVE pilot test wells (SVE-2, SVE-3, and SVE-4) at a range of levels (ranging from 34 to 115 inches of water vacuum); and (2) vacuum pressure was measured at five monitoring points (SVE-1, SVE-2, SVE-3, SVE-4, MW-3, and/or MW-9) at varying distances from the extraction well. Vacuum was induced using a self-contained mobile blower unit. For each phase, a constant flow rate was maintained and the vacuum pressure in the extraction well and the five monitoring wells was periodically measured and recorded until conditions stabilized. SVE pilot test well locations are shown on Figure 3; field notes and pilot test flow rate data are presented in Appendix A.

3.1.1 Estimated Radius of Influence

Using data gathered during the pilot test, an analytical model (e.g., Air2D; USGS, 1987) was used to estimate the intrinsic air permeability of the geologic formation at the Site (using the collected data from each monitoring point, at each flow rate, and at varying radial distances of each monitoring well from the active extraction point). The estimated intrinsic air permeability was determined to be approximately $1 \times 10^{-7}$ square centimeters ($\text{cm}^2$). Using this derived permeability, the Air2D model was also used to assess the site-specific relationship between
air flow (and the resultant vacuum pressure) and the potential radius of influence (ROI) of each soil vapor extraction point. The results of the Air2D model are included in Appendix A.

In accordance with the U.S. Army Corps of Engineers (USACE) design guidance for SVE systems (USACE, 2002), the ROI is defined as the distance at which the soil vapor velocity is a minimum of 0.001 cm/sec. Based on this evaluation, the ROI for an SVE system at the Site would be between 19 feet and 24 feet (corresponding to a per-well air flow of 7 to 10 cubic feet per minute [cfm]). Higher soil vapor flow rates may be attainable, but would be potentially unsustainable due to high stresses on system equipment.

3.1.2 VOC Concentrations in System Effluent

At intervals during the SVE pilot test, volatile organic compound (VOC) concentrations were measured at the blower exhaust using a photoionization detector (PID). In addition, two samples of the system effluent were collected and submitted for laboratory analysis of VOCs (benzene, 1,2-dichloroethane, ethylbenzene, naphthalene, toluene, 1,2,4-trimethylbenzene, 1,3,5-trimethylbenzene, xylenes, n-butylbenzene, and n-propylbenzene) using U.S. Environmental Protection Agency (EPA) Method TO-15.

As shown in Table 1, effluent samples collected during the SVE pilot study identified an overall VOC concentration in effluent of up to 50,000 micrograms per cubic meter (µg/m³). The analytical laboratory report is included in Appendix B. The effluent data were evaluated to assess whether effluent discharge from a conceptual SVE system would meet Kansas criteria for discharge to the atmosphere, or if effluent treatment would be required prior to discharge. The effluent evaluation is discussed in more detail in Section 3.2.

3.2 System Layout and Specifications

Based on the SVE pilot test data, presented in Section 3.1, and the understanding of the distribution of gasoline constituents in soil, described in detail in the CI report, an SVE system was designed that would maximize removal of hydrocarbon mass from the subsurface, while minimizing aesthetic impacts and inconvenience to the Neighborhood. The proposed location of the extraction well network for the conceptual SVE system is within the immediate release area, which contains the highest vadose-zone concentrations of gasoline constituents at the Site. The conceptual location for the aboveground components of the system is on community-owned land in proximity to the source area, maximizing the efficiency of the conceptual system. The SVE system design and specifications are presented below.

The layout and size of SVE system components is dependent on the surface cover, soil type, depth to groundwater (the thickness of the vadose zone), and horizontal and vertical extent of the target VOCs. This section provides the parameters used for the design of the SVE system. In general, the design is based on the recommendations provided in the USACE Soil Vapor Extraction Engineer Manual (2002) and the results of the December 2013 SVE pilot test.
The conceptual layout of the system is shown on Figure 4, a schematic of the SVE system components is shown on Figure 5, and the conceptual remediation building is shown on Figure 6. The bulk of the underground components of the SVE system (e.g., extraction wells) would be installed in the NuStar easement and City of Andover ROW in the vicinity of the pipeline release location. Aboveground infrastructure, consisting of a blower, control systems, knock-out drum, and effluent treatment vessels, would be housed in a small building constructed on property owned by the Quail Crossing Homeowner Association (HOA) south of the NuStar pipeline. The building would secure and weatherize aboveground equipment, improve equipment aesthetics, and mitigate noise from the blower.

**Chemical Profile.** The SVE system was designed to target gasoline constituents. Published information about the volatility of these gasoline constituents has been used to verify the applicability of SVE as a removal technology.

**Soil Profile.** The ground surface in the treatment area is mostly paved. The underlying soil profile consists of continuous fine-grained soil (clays) through the thickness of the vadose zone (to a typical depth of about 30 feet bgs). The intrinsic air permeability measured during the SVE pilot study is $5 \times 10^{-6}$ cm$^2$. The air permeability, calculated based on the SVE pilot test data, is provided in Appendix A.

**Extraction Points.** As discussed in Section 3.1, the SVE extraction wells would have an ROI of up to 24 feet. To provide complete coverage over the treatment area, the radii of each well should overlap by about 20 percent (so that the radii of adjacent offset rows converge). The proposed SVE well spacing for the conceptual design, therefore, is 38 feet. At this spacing, a total of 5 wells would be needed to remove source area material in the vicinity of the release location. The proposed layout of the vapor extraction points is shown on Figure 4. Each proposed extraction well would be comprised of a 2-inch-diameter Schedule 80 PVC well, with 10 feet of screen, installed between approximately 20 and 30 feet bgs.

**Blower Selection.** The required air flow for the SVE system is on the order of up to 50 cfm, based upon five SVE wells at a flow rate of 10 cfm each and vacuum pressures of on the order of 80 inches of water. These specifications are within the operating parameters of medium-sized positive displacement blowers (such as the 7.5 hp Rotron EN633).

**Piping Selection and Layout.** For the conceptual design, piping is sized so that the pressure drop through the pipe (due to friction losses) is less than 0.01 inch of water per foot of pipe. The selected piping diameters are 2 inches at each vapor extraction well and for lengths of single-well piping. Larger piping (3- to 4-inch-diameter) would be used for manifold piping and blower system components. The proposed piping consists of Schedule 80 PVC. The concentrations of vapors in the operating airflow are not expected to adversely affect the PVC material during the operating life of the system. The conceptual layout of the system is shown on Figure 5. Piping would be installed below ground.
Effluent Treatment. Activated carbon would be used to remove VOCs from the air stream prior to discharge into the atmosphere (other removal technologies, such as catalytic or thermal oxidation, require significant energy and are generally incompatible with the neighborhood noise levels). Two approximately 1,000-pound carbon vessels would be installed in series to treat the effluent. The carbon vessels are sized to accommodate the required air flow and to reasonably limit the frequency of carbon replacement. Effluent monitoring would be performed before, between, and after the carbon vessels to verify efficient use of the carbon and compliance with air discharge limitations.

Based on effluent measurements collected during the SVE pilot test, the estimated initial carbon usage rates are on the order of 30 pounds of carbon per day. While even small effluent carbon vessels (i.e., 55-gallon drums) will allow the design flow rate of 50 cfm (with a head loss of less than 5 inches of water), the mass loading on such a vessel would require replacement frequency that would be on the order of 6 days. To allow a reasonable frequency of carbon replacements (requiring mobilization to the site, removal of the spent carbon for regeneration, and replacement of new carbon), the system would be installed with two 1,000-pound carbon vessels, which would have an initial change-out frequency of on the order of monthly. It is expected that the usage rate would decrease with time as the VOC concentrations in the treatment area are reduced. The carbon usage rate would be re-evaluated after the startup of the system to assess the likely frequency of carbon changes.

Noise Mitigation. Noise generated by an SVE system includes mechanical noise generated by the blower unit and flow noise from air being discharged from the exhaust stack. Without mitigation, this noise can exceed 80 decibels (dba) in close proximity to the equipment. To reduce noise, the exhaust stack would be oversized to reduce air velocity and would include a muffler unit. The system equipment would be enclosed in an insulated building (as shown on Figure 6), which would reduce sound emissions to generally acceptable levels (i.e., less than 40 dba at a distance of 20 feet, a reference sound level consistent with quiet suburban neighborhoods). If further noise reduction is required (such as between the system and a sensitive receptor), a sound attenuation wall could be installed to further reduce noise levels.

3.3 System Operation

System Startup. Initial startup of the system includes powering up of the SVE blower and verifying proper operation of the equipment (including measurement of the induced vacuum at each of the SVE wellheads, monitoring wells, and at each of the system influents). Vacuum measurements at the influent are monitored to verify system performance (e.g., changes in vacuum pressure may indicate problems in system operation) and to verify that the blower is not overloaded (excessive input vacuums may overstress the blower). Vacuum pressures at each of the SVE wellheads would be balanced by adjusting flow control valves at each wellhead. The vacuum pressure would be monitored while the system is in operation, using magnehelic pressure gauges.
Vacuum pressure data would be evaluated to confirm the ROI of the vapor extraction wells. Adjustments to the system would be made as appropriate (such as adjusting the air flow rate) to provide operation consistent with the design goals.

**Startup Monitoring.** Following the initial startup of the system, VOCs would be measured at each wellhead and at blower effluent using a PID. The PID measures volatile organics in the effluent stream using a low vacuum to withdraw an effluent sample from the air stream. Therefore, samples from the vacuum side (the influent branches) of the blower are collected with the system off. Samples of the system effluent would also be collected and submitted for laboratory analysis of VOC concentrations.

The data from the startup monitoring would be used to assess the operation of the system and the carbon vessel mass loading rate. Carbon vessel sizing may be adjusted if actual mass removal rates are significantly different from the conditions encountered during the pilot test.

**Routine System Monitoring.** Routine monitoring of the SVE system would be conducted weekly during the first month of operation and monthly thereafter. Effluent sampling downstream of the first of the two carbon vessels would allow a determination of when the carbon has been saturated and breakthrough is occurring (with the second carbon maintaining protection of air quality). When breakthrough of the first (lead) carbon occurs, the spent carbon would be exchanged for fresh carbon and the carbon vessels would be reversed such that the new carbon is in the second (lag) position while the former lag carbon becomes the lead carbon. The monitoring consists of PID and pressure monitoring. System effluent samples will be collected to corroborate the PID data. The results of ongoing system monitoring may be used to adjust the monitoring frequency if appropriate.

### 3.4 Schedule

Implementing the system described herein could begin within six months of KDHE and Neighborhood approval, barring delays beyond the control of NuStar.

### 4.0 Neighborhood Meeting and Public Outreach

On February 17, 2014, a public meeting was held near the Quail Crossing Neighborhood. Attendees included approximately 50 Neighborhood residents; as well as representatives from KDHE, NuStar, and Apex. NuStar representatives presented to the residents the results of the CI followed by the proposed SVE design for remediation. The suburban residential setting of the Site was taken into consideration and played a key role in the conceptual cleanup design. As such, the proposal to the residents highlighted design enhancements and modifications including the location of infrastructure, noise mitigation, and aesthetics.

- The extraction well network will be located primarily in W Mountain Street in the immediate release area adjacent to the pipeline rather than on private property.
- The conceptual remediation building, as shown on Figure 6, was enhanced to be more aesthetically pleasing, with an appearance more similar to other structures in the Neighborhood rather than standard remediation design.

- Sound dampening insulation was proposed for use in the building to minimize the decibel level. In addition, a separate sound attenuation wall was proposed outside of the building to further reduce noise levels.

- The proposed location for the remediation building was on HOA property rather than private homeowner property.

- Fencing and/or a wall along the abutting property boundary was proposed for additional aesthetic and noise mitigation.

Responses during the meeting and following the meeting indicated that Neighborhood residents were overwhelmingly opposed to construction and operation of an SVE system in the Neighborhood. The primary concerns cited as a basis for opposition were: (1) sound levels associated with operation of the SVE system; (2) visual impact of the system enclosure; and (3) the expected duration of operation of an SVE system. In an email dated February 22, 2014, the Quail Crossing Neighborhood Association stated, “The proposed building site is unacceptable because it is too close to the Lee’s home [1023 W Mountain] and because it will significantly detract from and negatively impact the Lee’s home and the neighborhood’s appearance.”

NuStar and KDHE reviewed the feedback provided by Neighborhood residents and, in consideration of the concerns voiced by the residents, determined that SVE was not a feasible approach for removing petroleum hydrocarbon mass from the subsurface. In the email dated February 22, 2014, the Quail Crossing Neighborhood Association stated that the residents wanted to evaluate other source removal options, including excavation of the soil that contains gasoline constituents. In response to the Neighborhood request, and as outlined in the next section, NuStar proposes to perform additional soil investigation to refine the understanding of the extent of soil with gasoline constituents and to evaluate excavation as a source removal option for the Site in lieu of SVE and/or other remedial technologies which would require similar infrastructure.

5.0 Soil Investigation

The following sections detail the methods for the proposed soil investigation at the Site. The soil investigation will be focused on the pipeline release area, where residual gasoline constituents are present in vadose zone soil. The information obtained during the soil investigation will be used to evaluate alternative source removal options for the Site, including soil excavation.
5.1 Proposed Soil Boring Locations

In June 2012, NuStar excavated approximately 16 cubic yards of soil containing gasoline constituents from the pipeline release area. The lateral extent of the excavation was limited by buried utilities and private property. Soil data obtained during the CI indicate that soil containing gasoline constituents is generally limited to the immediate vicinity of the pipeline release location. Consequently, twelve soil borings are proposed in this area (within approximately 15 feet of the NuStar pipeline) to further delineate the lateral and vertical extent of gasoline constituents in soil. The proposed locations for the soil borings are shown on Figure 7. The number and locations of soil borings may be adjusted based on field observations during drilling.

5.2 Preparatory Activities

Preparatory activities, such as the preparation of a Health and Safety Plan (HASP), procurement of subcontractors, site clearing, and the location of underground utilities, will be performed. These activities are detailed in the Sampling and Analysis Plan (SAP) in Appendix C.

5.3 Procedures

The specific procedures for completing field work are detailed in the SAP in Appendix C. The following provides a brief description of the methods to be used for the soil investigation.

5.4 Drilling and Soil Sample Collection

Borings will be advanced using a hollow stem auger (HSA) drill rig. Borings will be advanced to an approximate depth of 20 feet bgs. Soil cores will be extracted and soil lithology and conditions will be logged in general accordance with ASTM 2487/2488. Descriptions will include visual indications of petroleum impacts. Soil cores will be field screened for VOCs and SPH using a PID, and sheen tests will be conducted on soil at 2.5-foot intervals. Detailed soil sampling procedures and the proposed analytical program are described in Appendix C. A field log of lithology and soil conditions will be maintained for the borings. Drilling equipment (i.e., auger stems, split-spoon sampler, etc.) will be decontaminated before and after each boring is complete. The location of each boring will be recorded using a handheld global positioning system (GPS) instrument.

5.5 Soil Sampling and Analysis

Soil samples will be collected using split-spoon sampling equipment. The split-spoon sampling method provides essentially continuous soil samples during the advancement of the boring. Discrete soil samples will be collected at approximately 2.5-foot intervals using the split spoon. Each soil sample will be logged for lithologic description, field screened, and held for possible chemical analysis. Samples will be selected for laboratory analysis based on field screening results, and as necessary to adequately define the nature and
extent of gasoline constituents in the subsurface. If the soil type changes or field screening suggests contamination in a portion of the core, each distinct portion of the core will be sampled. Up to six soil samples from each boring will be submitted for laboratory analysis, including one sample from the bottom of each boring. Detailed soil sampling procedures are described in the SAP (Appendix C).

Soil samples will be submitted to an accredited laboratory for analysis of: gasoline-range organics (GRO) by Iowa Method OA-1; and benzene, toluene, ethylbenzene, and xylenes (BTEX), 2-butanone, n-buty1benzene, naphthalene, n-propylbenzene, 1,2,4-trimethylbenzene, 1,3,5-trimethylbenzene, cyclohexane, isopropylbenzene, and methylcyclohexane by EPA Method 8260B.

5.6 Field Screening

During the completion of the borings, soil cores will be screened for VOCs using a PID and SPH through sheen testing. The field screening procedures are outlined in the SAP, which is included as Appendix C.

5.7 Quality Assurance and Quality Control

Quality assurance / quality control (QA/QC) procedures will be used throughout this project and are described in detail in the SAP (Appendix C). The QA/QC procedures detailed in the SAP include sampling and custody procedures, QA sampling analyses (such as analysis of duplicates), laboratory detection limit goals, laboratory QC, and QA reporting.

5.8 Investigation-Derived Waste

Investigation-derived waste (IDW) will consist of soil from the soil cores, decontamination water, and personal protection equipment (PPE). Soil and water IDW be containerized, labeled with the project name, general contents, and date. The IDW will be profiled and transported to appropriate facilities for disposal.

6.0 Schedule

The scope of work described herein will be initiated within 60 days of KDHE’s approval of the Work Plan, barring delays beyond the control of NuStar. Access agreements required to advance borings and stage drilling equipment were finalized prior to performance of the Comprehensive Investigation in 2013. It is estimated that the field activities will take approximately three days. The results of the investigation will be presented and discussed in a report to be submitted within 60 days following receipt of the analytical data. The report will: (1) document the results of the activities proposed herein; and (2) present an evaluation of the analytical data. Based on the analytical data review, NuStar may propose excavation of soil that contains gasoline constituents, or other appropriate measures.
If additional investigation is necessary to define the nature and extent soil with gasoline constituents following implementation of the work described herein, NuStar will coordinate with KDHE regarding follow-up activities and schedule.

7.0 References


### Table 1
**SVE Pilot Test Effluent Data**
**Andover Colt Court Release Site**
**Andover, Kansas**

<table>
<thead>
<tr>
<th>Sample Location</th>
<th>Sample Date</th>
<th>1,2 Dichloroethane</th>
<th>Benzene</th>
<th>Toluene</th>
<th>Ethylbenzene</th>
<th>Total Xylenes</th>
<th>n-Propylbenzene</th>
<th>1,3,5 Trimethylbenzene</th>
<th>1,2,4 Trimethylbenzene</th>
<th>Naphthalene</th>
<th>n-Butylbenzene</th>
</tr>
</thead>
<tbody>
<tr>
<td>SVE Initial</td>
<td>12/10/2013</td>
<td>&lt;2,500</td>
<td>11,000</td>
<td>24,000</td>
<td>&lt;2,500</td>
<td>&lt;7,500</td>
<td>&lt;2,500</td>
<td>&lt;2,500</td>
<td>&lt;2,500</td>
<td>&lt;2,500</td>
<td>&lt;2,500</td>
</tr>
<tr>
<td>SVE Completion</td>
<td>12/10/2013</td>
<td>&lt;830</td>
<td>3,200</td>
<td>18,000</td>
<td>5,800</td>
<td>&lt;830</td>
<td>&lt;830</td>
<td>20,700</td>
<td>&lt;830</td>
<td>1,000</td>
<td>&lt;830</td>
</tr>
</tbody>
</table>

**Notes:**
1. 1,2-dichloroethane, benzene, toluene, ethylbenzene, total xylenes, n-propylbenzene, 1,3,5 trimethylbenzene, 1,2,4 trimethylbenzene, naphthalene, and n-butylbenzene by Environmental Protection Agency (EPA) Method TO-15.
2. ug/m³ (ppm) = Micrograms per cubic meter.
3. < = Not detected above the method reporting limit (MRL).
Note: Base map prepared from USGS 7.5-minute quadrangles of Andover and Santa Fe Lake, KS, dated 2009 as provided by USGS.gov.
Soil Vapor Extraction Test Well Locations

Legend:
- MW-1 Monitoring Well Location
- Irrigation Well
- SVE-1 Soil Vapor Extraction Pilot Test Well
- Property Line
- Pipeline
- Pipeline Easement Boundary
- Pipeline Excavation Extent (June 14, 2012)
- HOA Lot Owned by Quail Crossing Homeowner’s Association

Soil Vapor Extraction Evaluation and Proposed Soil Investigation Report
NuStar Pipeline Operating Partnership L.P.
Andover, Kansas

Apex Companies, LLC
3015 SW First Avenue
Portland, Oregon 97201

Project Number 1641-04
Figure 3
March 2014
Legend:
- SVE Well - Soil Vapor Extraction (SVE) Well Location
- SVE-3 - SVE Well Location (December 2013)
- SVE Piping
- Anticipated 20' Radius of Vacuum Influence
- MW-3 - Monitoring Well Location
- Irrigation Well
- B-1 - Boring Location
- Floor - Soil Sample Location
- Property Line
- Pipeline
- Pipeline Easement Boundary
- Pipeline Excavation Extent (July 5-7, 2012 and June 14, 2012)
- HOA - Lot Owned by Quail Crossing Homeowner's Association
- Sample Depth (Feet)
- Benzene Concentration in mg/kg (Highlight Indicates a Concentration Above KDHE Tier 2 RBSV for Soil to Groundwater Pathway; 0.168 mg/kg)
- NS - Not Sampled

Conceptual SVE System
Soil Vapor Extraction Evaluation and Proposed Soil Investigation Report
NuStar Pipeline Operating Partnership L.P.
Andover, Kansas

Aerial photograph provided by Google Maps.com (dated February 25, 2012).
Proposed SVE Enclosure

Soil Vapor Extraction Evaluation and Proposed Soil Investigation Report
NuStar Pipeline Operating Partnership L.P.
Andover, Kansas
Legend:
- Proposed Soil Boring Location
- Monitoring Well Location
- Irrigation Well
- Boring Location
- Soil Sample Location
- Property Line
- Pipeline
- Pipeline Easement Boundary
- Pipeline Excavation Extent (July 5-7, 2012 and June 14, 2012)
- Sample Depth (Feet)
- Benzene Concentration in mg/kg
  - Highlight Indicates a Concentration Above KDHE Tier 2 RBSV for Soil to Groundwater Pathway; 0.168 mg/kg
- Not Sampled

Proposed Soil Boring Locations

Soil Vapor Extraction Evaluation and Proposed Soil Investigation Report
NuStar Pipeline Operating Partnership L.P.
Andover, Kansas

Excavation (July 5-7, 2012)
Excavation (June 14, 2012)

Aerial photograph provided by Google Maps.com (dated February 25, 2012).
Appendix A

Field Notes and SVE Pilot Test Flow Rate Data
VOLUMETRIC FLOW SIGN CONVENTION:

NEGATIVE HORIZONTAL FLOW => FLOW TOWARDS WELL

POSITIVE HORIZONTAL FLOW => FLOW AWAY FROM WELL

NEGATIVE VERTICAL FLOW => FLOW UPWARDS

POSITIVE VERTICAL FLOW => FLOW DOWNWARDS
<table>
<thead>
<tr>
<th>Time (s)</th>
<th>x (m)</th>
<th>y (m)</th>
<th>Volumetric Flow Rate (L/s)</th>
<th>Mass Flow Rate (kg/s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>152.40</td>
<td>-731.52</td>
<td>-0.643413E-02</td>
<td>0.615366E-04</td>
<td>-0.643442E-02</td>
</tr>
<tr>
<td>182.88</td>
<td>-731.52</td>
<td>-0.519663E-02</td>
<td>0.220581E-03</td>
<td>-0.520131E-02</td>
</tr>
<tr>
<td>213.36</td>
<td>-731.52</td>
<td>-0.431097E-02</td>
<td>0.320858E-03</td>
<td>-0.432289E-02</td>
</tr>
<tr>
<td>243.84</td>
<td>-731.52</td>
<td>-0.364661E-02</td>
<td>0.377523E-03</td>
<td>-0.366610E-02</td>
</tr>
<tr>
<td>274.32</td>
<td>-731.52</td>
<td>-0.313068E-02</td>
<td>0.403817E-03</td>
<td>-0.315662E-02</td>
</tr>
<tr>
<td>304.80</td>
<td>-731.52</td>
<td>-0.271925E-02</td>
<td>0.409889E-03</td>
<td>-0.274997E-02</td>
</tr>
<tr>
<td>335.28</td>
<td>-731.52</td>
<td>-0.238419E-02</td>
<td>0.403028E-03</td>
<td>-0.241802E-02</td>
</tr>
<tr>
<td>365.76</td>
<td>-731.52</td>
<td>-0.210666E-02</td>
<td>0.388266E-03</td>
<td>-0.214215E-02</td>
</tr>
<tr>
<td>396.24</td>
<td>-731.52</td>
<td>-0.187355E-02</td>
<td>0.369022E-03</td>
<td>-0.190955E-02</td>
</tr>
<tr>
<td>426.72</td>
<td>-731.52</td>
<td>-0.167543E-02</td>
<td>0.347554E-03</td>
<td>-0.171110E-02</td>
</tr>
<tr>
<td>457.20</td>
<td>-731.52</td>
<td>-0.150537E-02</td>
<td>0.325325E-03</td>
<td>-0.154013E-02</td>
</tr>
<tr>
<td>487.68</td>
<td>-731.52</td>
<td>-0.135814E-02</td>
<td>0.303260E-03</td>
<td>-0.139159E-02</td>
</tr>
<tr>
<td>518.16</td>
<td>-731.52</td>
<td>-0.122973E-02</td>
<td>0.281924E-03</td>
<td>-0.126163E-02</td>
</tr>
<tr>
<td>548.64</td>
<td>-731.52</td>
<td>-0.111699E-02</td>
<td>0.261634E-03</td>
<td>-0.114722E-02</td>
</tr>
<tr>
<td>579.12</td>
<td>-731.52</td>
<td>-0.101745E-02</td>
<td>0.242579E-03</td>
<td>-0.104597E-02</td>
</tr>
<tr>
<td>609.60</td>
<td>-731.52</td>
<td>-0.0929116E-03</td>
<td>0.224806E-03</td>
<td>-0.095592E-03</td>
</tr>
<tr>
<td>640.08</td>
<td>-731.52</td>
<td>-0.0850366E-03</td>
<td>0.208318E-03</td>
<td>-0.087551E-03</td>
</tr>
<tr>
<td>670.56</td>
<td>-731.52</td>
<td>-0.0779876E-03</td>
<td>0.193076E-03</td>
<td>-0.080342E-03</td>
</tr>
<tr>
<td>701.04</td>
<td>-731.52</td>
<td>-0.0716548E-03</td>
<td>0.179018E-03</td>
<td>-0.073857E-03</td>
</tr>
<tr>
<td>731.52</td>
<td>-731.52</td>
<td>-0.0659466E-03</td>
<td>0.166068E-03</td>
<td>-0.068005E-03</td>
</tr>
<tr>
<td>762.00</td>
<td>-731.52</td>
<td>-0.0607857E-03</td>
<td>0.154148E-03</td>
<td>-0.062709E-03</td>
</tr>
<tr>
<td>792.48</td>
<td>-731.52</td>
<td>-0.0561071E-03</td>
<td>0.143178E-03</td>
<td>-0.057905E-03</td>
</tr>
<tr>
<td>822.96</td>
<td>-731.52</td>
<td>-0.0518549E-03</td>
<td>0.133080E-03</td>
<td>-0.053535E-03</td>
</tr>
<tr>
<td>853.44</td>
<td>-731.52</td>
<td>-0.0479816E-03</td>
<td>0.123780E-03</td>
<td>-0.049525E-03</td>
</tr>
<tr>
<td>883.92</td>
<td>-731.52</td>
<td>-0.0444458E-03</td>
<td>0.115211E-03</td>
<td>-0.045914E-03</td>
</tr>
</tbody>
</table>

**VOLUMETRIC FLOW SIGN CONVENTION**

---------- ---- ---- ----------

NEGATIVE HORIZONTAL FLOW => FLOW TOWARDS WELL
POSITIVE HORIZONTAL FLOW => FLOW AWAY FROM WELL
NEGATIVE VERTICAL FLOW  => FLOW UPWARDS
POSITIVE VERTICAL FLOW  => FLOW DOWNWARDS
<table>
<thead>
<tr>
<th>Time</th>
<th>X-Y-Z</th>
<th>Volumetric Flow</th>
<th>Horizontal Flow</th>
<th>Vertical Flow</th>
</tr>
</thead>
<tbody>
<tr>
<td>152.40</td>
<td>-731.52</td>
<td>-0.968668E-02</td>
<td>0.926443E-04</td>
<td>-0.968712E-02</td>
</tr>
<tr>
<td>182.88</td>
<td>-731.52</td>
<td>-0.782018E-02</td>
<td>0.331944E-03</td>
<td>-0.782723E-02</td>
</tr>
<tr>
<td>213.36</td>
<td>-731.52</td>
<td>-0.648508E-02</td>
<td>0.482674E-03</td>
<td>-0.650302E-02</td>
</tr>
<tr>
<td>243.84</td>
<td>-731.52</td>
<td>-0.548404E-02</td>
<td>0.567747E-03</td>
<td>-0.551335E-02</td>
</tr>
<tr>
<td>274.32</td>
<td>-731.52</td>
<td>-0.470696E-02</td>
<td>0.607136E-03</td>
<td>-0.474596E-02</td>
</tr>
<tr>
<td>304.80</td>
<td>-731.52</td>
<td>-0.408748E-02</td>
<td>0.616131E-03</td>
<td>-0.413366E-02</td>
</tr>
<tr>
<td>335.28</td>
<td>-731.52</td>
<td>-0.358315E-02</td>
<td>0.605703E-03</td>
<td>-0.363399E-02</td>
</tr>
<tr>
<td>365.76</td>
<td>-731.52</td>
<td>-0.316553E-02</td>
<td>0.583419E-03</td>
<td>-0.321885E-02</td>
</tr>
<tr>
<td>396.24</td>
<td>-731.52</td>
<td>-0.281483E-02</td>
<td>0.554421E-03</td>
<td>-0.286891E-02</td>
</tr>
<tr>
<td>426.72</td>
<td>-731.52</td>
<td>-0.251685E-02</td>
<td>0.522099E-03</td>
<td>-0.257043E-02</td>
</tr>
<tr>
<td>457.20</td>
<td>-731.52</td>
<td>-0.226112E-02</td>
<td>0.488648E-03</td>
<td>-0.231331E-02</td>
</tr>
<tr>
<td>487.68</td>
<td>-731.52</td>
<td>-0.203976E-02</td>
<td>0.455457E-03</td>
<td>-0.208999E-02</td>
</tr>
<tr>
<td>518.16</td>
<td>-731.52</td>
<td>-0.184671E-02</td>
<td>0.423373E-03</td>
<td>-0.189462E-02</td>
</tr>
<tr>
<td>548.64</td>
<td>-731.52</td>
<td>-0.167727E-02</td>
<td>0.392869E-03</td>
<td>-0.172267E-02</td>
</tr>
<tr>
<td>579.12</td>
<td>-731.52</td>
<td>-0.152768E-02</td>
<td>0.364226E-03</td>
<td>-0.157050E-02</td>
</tr>
<tr>
<td>609.60</td>
<td>-731.52</td>
<td>-0.139495E-02</td>
<td>0.337517E-03</td>
<td>-0.143520E-02</td>
</tr>
<tr>
<td>640.08</td>
<td>-731.52</td>
<td>-0.127663E-02</td>
<td>0.312742E-03</td>
<td>-0.131438E-02</td>
</tr>
<tr>
<td>670.56</td>
<td>-731.52</td>
<td>-0.117074E-02</td>
<td>0.289842E-03</td>
<td>-0.120608E-02</td>
</tr>
<tr>
<td>701.04</td>
<td>-731.52</td>
<td>-0.107561E-02</td>
<td>0.268723E-03</td>
<td>-0.110867E-02</td>
</tr>
<tr>
<td>731.52</td>
<td>-731.52</td>
<td>-0.989872E-03</td>
<td>0.249272E-03</td>
<td>-0.102078E-02</td>
</tr>
<tr>
<td>762.00</td>
<td>-731.52</td>
<td>-0.912364E-03</td>
<td>0.231369E-03</td>
<td>-0.941244E-03</td>
</tr>
<tr>
<td>792.48</td>
<td>-731.52</td>
<td>-0.842103E-03</td>
<td>0.214893E-03</td>
<td>-0.869090E-03</td>
</tr>
<tr>
<td>822.96</td>
<td>-731.52</td>
<td>-0.778252E-03</td>
<td>0.199729E-03</td>
<td>-0.803473E-03</td>
</tr>
<tr>
<td>853.44</td>
<td>-731.52</td>
<td>-0.720993E-03</td>
<td>0.185766E-03</td>
<td>-0.743669E-03</td>
</tr>
<tr>
<td>883.92</td>
<td>-731.52</td>
<td>-0.667007E-03</td>
<td>0.172900E-03</td>
<td>-0.689052E-03</td>
</tr>
</tbody>
</table>

**VOLUMETRIC FLOW SIGN CONVENTION:**

---

**NEGATIVE HORIZONTAL FLOW** => FLOW TOWARDS WELL

**POSITIVE HORIZONTAL FLOW** => FLOW AWAY FROM WELL

**NEGATIVE VERTICAL FLOW**  => FLOW UPWARDS

**POSITIVE VERTICAL FLOW**  => FLOW DOWNWARDS
<table>
<thead>
<tr>
<th>Time (s)</th>
<th>Distance (m)</th>
<th>Velocity (m/s)</th>
<th>Acceleration (m/s²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>152.40</td>
<td>-731.52</td>
<td>-0.129634E-01</td>
<td>-0.129640E-01</td>
</tr>
<tr>
<td>182.88</td>
<td>-731.52</td>
<td>-0.104609E-01</td>
<td>-0.104703E-01</td>
</tr>
<tr>
<td>213.36</td>
<td>-731.52</td>
<td>-0.867183E-02</td>
<td>-0.869582E-02</td>
</tr>
<tr>
<td>243.84</td>
<td>-731.52</td>
<td>-0.733103E-02</td>
<td>-0.737021E-02</td>
</tr>
<tr>
<td>274.32</td>
<td>-731.52</td>
<td>-0.629063E-02</td>
<td>-0.634275E-02</td>
</tr>
<tr>
<td>304.80</td>
<td>-731.52</td>
<td>-0.546153E-02</td>
<td>-0.552323E-02</td>
</tr>
<tr>
<td>335.28</td>
<td>-731.52</td>
<td>-0.478675E-02</td>
<td>-0.485465E-02</td>
</tr>
<tr>
<td>365.76</td>
<td>-731.52</td>
<td>-0.422813E-02</td>
<td>-0.429934E-02</td>
</tr>
<tr>
<td>396.24</td>
<td>-731.52</td>
<td>-0.375915E-02</td>
<td>-0.383137E-02</td>
</tr>
<tr>
<td>426.72</td>
<td>-731.52</td>
<td>-0.336075E-02</td>
<td>-0.343230E-02</td>
</tr>
<tr>
<td>457.20</td>
<td>-731.52</td>
<td>-0.301891E-02</td>
<td>-0.308861E-02</td>
</tr>
<tr>
<td>487.68</td>
<td>-731.52</td>
<td>-0.272307E-02</td>
<td>-0.279013E-02</td>
</tr>
<tr>
<td>518.16</td>
<td>-731.52</td>
<td>-0.246513E-02</td>
<td>-0.252908E-02</td>
</tr>
<tr>
<td>548.64</td>
<td>-731.52</td>
<td>-0.223875E-02</td>
<td>-0.229934E-02</td>
</tr>
<tr>
<td>579.12</td>
<td>-731.52</td>
<td>-0.203892E-02</td>
<td>-0.209607E-02</td>
</tr>
<tr>
<td>609.60</td>
<td>-731.52</td>
<td>-0.186163E-02</td>
<td>-0.191535E-02</td>
</tr>
<tr>
<td>640.08</td>
<td>-731.52</td>
<td>-0.170362E-02</td>
<td>-0.175399E-02</td>
</tr>
<tr>
<td>670.56</td>
<td>-731.52</td>
<td>-0.156221E-02</td>
<td>-0.160937E-02</td>
</tr>
<tr>
<td>701.04</td>
<td>-731.52</td>
<td>-0.143520E-02</td>
<td>-0.147931E-02</td>
</tr>
<tr>
<td>731.52</td>
<td>-731.52</td>
<td>-0.132073E-02</td>
<td>-0.136196E-02</td>
</tr>
<tr>
<td>762.00</td>
<td>-731.52</td>
<td>-0.121726E-02</td>
<td>-0.125579E-02</td>
</tr>
<tr>
<td>792.48</td>
<td>-731.52</td>
<td>-0.112347E-02</td>
<td>-0.115947E-02</td>
</tr>
<tr>
<td>822.96</td>
<td>-731.52</td>
<td>-0.103824E-02</td>
<td>-0.107189E-02</td>
</tr>
<tr>
<td>853.44</td>
<td>-731.52</td>
<td>-0.960619E-03</td>
<td>-0.992069E-03</td>
</tr>
<tr>
<td>883.92</td>
<td>-731.52</td>
<td>-0.889770E-03</td>
<td>-0.919178E-03</td>
</tr>
</tbody>
</table>

VOLUMETRIC FLOW SIGN CONVENTION:
-------------- ---- ---- ------------
NEGATIVE HORIZONTAL FLOW => FLOW TOWARDS WELL
POSITIVE HORIZONTAL FLOW => FLOW AWAY FROM WELL
NEGATIVE VERTICAL FLOW  => FLOW UPWARDS
POSITIVE VERTICAL FLOW  => FLOW DOWNWARDS
<table>
<thead>
<tr>
<th>Time</th>
<th>Temperature</th>
<th>Volumetric Flow</th>
<th>Conventional Flow</th>
</tr>
</thead>
<tbody>
<tr>
<td>152.40</td>
<td>-731.52</td>
<td>-0.162647E-01</td>
<td>-0.162655E-01</td>
</tr>
<tr>
<td>182.88</td>
<td>-731.52</td>
<td>-0.131190E-01</td>
<td>-0.131308E-01</td>
</tr>
<tr>
<td>213.36</td>
<td>-731.52</td>
<td>-0.108714E-01</td>
<td>-0.109015E-01</td>
</tr>
<tr>
<td>243.84</td>
<td>-731.52</td>
<td>-0.091877E-01</td>
<td>-0.092368E-01</td>
</tr>
<tr>
<td>274.32</td>
<td>-731.52</td>
<td>-0.0788178E-02</td>
<td>-0.0794707E-02</td>
</tr>
<tr>
<td>304.80</td>
<td>-731.52</td>
<td>-0.0684144E-02</td>
<td>-0.0691872E-02</td>
</tr>
<tr>
<td>335.28</td>
<td>-731.52</td>
<td>-0.0599501E-02</td>
<td>-0.0608006E-02</td>
</tr>
<tr>
<td>365.76</td>
<td>-731.52</td>
<td>-0.0529449E-02</td>
<td>-0.0538366E-02</td>
</tr>
<tr>
<td>396.24</td>
<td>-731.52</td>
<td>-0.0470652E-02</td>
<td>-0.0479695E-02</td>
</tr>
<tr>
<td>426.72</td>
<td>-731.52</td>
<td>-0.0420716E-02</td>
<td>-0.0429673E-02</td>
</tr>
<tr>
<td>457.20</td>
<td>-731.52</td>
<td>-0.0377878E-02</td>
<td>-0.0386601E-02</td>
</tr>
<tr>
<td>487.68</td>
<td>-731.52</td>
<td>-0.0340811E-02</td>
<td>-0.0349204E-02</td>
</tr>
<tr>
<td>518.16</td>
<td>-731.52</td>
<td>-0.0308497E-02</td>
<td>-0.0316501E-02</td>
</tr>
<tr>
<td>548.64</td>
<td>-731.52</td>
<td>-0.0280142E-02</td>
<td>-0.0287725E-02</td>
</tr>
<tr>
<td>579.12</td>
<td>-731.52</td>
<td>-0.0255117E-02</td>
<td>-0.0262268E-02</td>
</tr>
<tr>
<td>609.60</td>
<td>-731.52</td>
<td>-0.0232917E-02</td>
<td>-0.0239638E-02</td>
</tr>
<tr>
<td>640.08</td>
<td>-731.52</td>
<td>-0.0213133E-02</td>
<td>-0.0219435E-02</td>
</tr>
<tr>
<td>670.56</td>
<td>-731.52</td>
<td>-0.0195430E-02</td>
<td>-0.0201331E-02</td>
</tr>
<tr>
<td>701.04</td>
<td>-731.52</td>
<td>-0.0179531E-02</td>
<td>-0.0185049E-02</td>
</tr>
<tr>
<td>731.52</td>
<td>-731.52</td>
<td>-0.0165204E-02</td>
<td>-0.0170361E-02</td>
</tr>
<tr>
<td>762.00</td>
<td>-731.52</td>
<td>-0.0152254E-02</td>
<td>-0.0157073E-02</td>
</tr>
<tr>
<td>792.48</td>
<td>-731.52</td>
<td>-0.0140517E-02</td>
<td>-0.0145020E-02</td>
</tr>
<tr>
<td>822.96</td>
<td>-731.52</td>
<td>-0.0129852E-02</td>
<td>-0.0134060E-02</td>
</tr>
<tr>
<td>853.44</td>
<td>-731.52</td>
<td>-0.0120139E-02</td>
<td>-0.0124072E-02</td>
</tr>
<tr>
<td>883.92</td>
<td>-731.52</td>
<td>-0.0111275E-02</td>
<td>-0.0114952E-02</td>
</tr>
</tbody>
</table>

**VOLUMETRIC FLOW SIGN CONVENTION:**

---------- ---- ---- ----------

NEGATIVE HORIZONTAL FLOW => FLOW TOWARDS WELL

POSITIVE HORIZONTAL FLOW => FLOW AWAY FROM WELL

NEGATIVE VERTICAL FLOW => FLOW UPWARDS

POSITIVE VERTICAL FLOW => FLOW DOWNWARDS
PROJECT : NUSTAR ANDOVER

SCOPE : RESULTS OF FULL-SCALE PERMEABILITY TESTS
TEST DATE : 12/10/2013
WELL NUMBER : SVE-2

1. MODEL INPUT SUMMARY
MODEL DOMAIN : THICKNESS = 914.400 cm
: ESTIMATED PERMEABILITY = 0.100E-07 cm^2
: ESTIMATED ANISOTROPY RATIO = 1.00
WELL DEPTH (HANTUSH d) : TOP OF SCREEN = 457.20 cm
WELL DEPTH (HANTUSH l) : BOTTOM OF SCREEN = 863.19 cm
WELL RADIUS : EFFECTIVE RADIUS = 15.24 cm
AIR FLOW DIRECTION : VAPOR EXTRACTION

2. MODEL OUTPUT SUMMARY

AIR SOIL ATMOS. SYSTEM FLOW SCALE PREVAIL. ACTUAL
TEMP TEMP PRESS. PRESS. METER READING FLOW FLOW
degC degC atm atm TYPE -- cm^3/sec cm^3/sec

<table>
<thead>
<tr>
<th>TEMP</th>
<th>TEMP</th>
<th>PRESS.</th>
<th>PRESS.</th>
<th>METER READING</th>
<th>FLOW</th>
<th>FLOW</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.00</td>
<td>15.00</td>
<td>1.000</td>
<td>0.727</td>
<td>NONE</td>
<td>--</td>
<td>3586.800</td>
</tr>
<tr>
<td>0.00</td>
<td>15.00</td>
<td>1.000</td>
<td>0.765</td>
<td>NONE</td>
<td>--</td>
<td>3303.630</td>
</tr>
</tbody>
</table>

MASS HORIZON. VERTICAL LEAKAGE ANISOTPY MEAN OF STD DEV
FLOW PERM. PERM. RATIO(k/b) RATIO OF ERROR IN OF ERROR
g/sec cm^2 cm^2 cm^2/cm (kr/kz) PRESS. IN PRESS.

<table>
<thead>
<tr>
<th>MASS</th>
<th>HORIZON</th>
<th>VERTICAL LEAKAGE</th>
<th>ANISOTPY</th>
<th>MEAN OF STD</th>
<th>DEV</th>
</tr>
</thead>
<tbody>
<tr>
<td>3.261</td>
<td>0.990E-07</td>
<td>0.990E-07</td>
<td>0.000E+00</td>
<td>1.000</td>
<td>-0.752E-03</td>
</tr>
<tr>
<td>3.161</td>
<td>0.277E-06</td>
<td>0.277E-06</td>
<td>0.000E+00</td>
<td>1.000</td>
<td>-0.783E-03</td>
</tr>
</tbody>
</table>
PROJECT : NUSTAR ANDOVER

SCOPE : RESULTS OF FULL-SCALE PERMEABILITY TESTS
TEST DATE : 12/10/2013
WELL NUMBER : SVE-4

1. MODEL INPUT SUMMARY

MODEL DOMAIN : THICKNESS = 914.400 cm
 : ESTIMATED PERMEABILITY = 0.900E-07 cm²
 : ESTIMATED ANISOTROPY RATIO = 1.00
WELL DEPTH (HANTUSH d) : TOP OF SCREEN = 609.60 cm
WELL DEPTH (HANTUSH l) : BOTTOM OF SCREEN = 882.70 cm
WELL RADIUS : EFFECTIVE RADIUS = 15.52 cm
AIR FLOW DIRECTION : VAPOR EXTRACTION

2. MODEL OUTPUT SUMMARY

AIR SOIL ATMOS. SYSTEM FLOW SCALE PREVAIL. ACTUAL
TEMP TEMP PRESS. PRESS. METER READING FLOW FLOW
degC degC atm atm TYPE -- cm³/sec cm³/sec

<table>
<thead>
<tr>
<th>AIR</th>
<th>SOIL</th>
<th>ATMOS.</th>
<th>SYSTEM</th>
<th>FLOW SCALE</th>
<th>PREVAIL.</th>
<th>ACTUAL</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.00</td>
<td>15.00</td>
<td>1.000</td>
<td>0.914</td>
<td>NONE</td>
<td>--</td>
<td>3303.630</td>
</tr>
<tr>
<td>0.00</td>
<td>15.00</td>
<td>1.000</td>
<td>0.864</td>
<td>NONE</td>
<td>--</td>
<td>5191.420</td>
</tr>
<tr>
<td>0.00</td>
<td>15.00</td>
<td>1.000</td>
<td>0.813</td>
<td>NONE</td>
<td>--</td>
<td>7551.160</td>
</tr>
</tbody>
</table>

MASS HORIZON. VERTICAL LEAKAGE ANISOTPY MEAN OF STD DEV
FLOW PERM. PERM. RATIO(k/b) RATIO ERROR IN OF ERROR
g/sec cm² cm² cm²/cm (kr/kz) PRESS. IN PRESS.

<table>
<thead>
<tr>
<th>MASS</th>
<th>HORIZON</th>
<th>VERTICAL LEAKAGE</th>
<th>ANISOTPY</th>
<th>MEAN OF STD DEV</th>
</tr>
</thead>
<tbody>
<tr>
<td>3.776</td>
<td>0.490E-07</td>
<td>0.490E-07</td>
<td>0.000E+00</td>
<td>1.000 0.231E-03</td>
</tr>
<tr>
<td>5.609</td>
<td>0.308E-07</td>
<td>0.308E-07</td>
<td>0.000E+00</td>
<td>1.000 0.152E-03</td>
</tr>
<tr>
<td>7.677</td>
<td>0.264E-07</td>
<td>0.264E-07</td>
<td>0.000E+00</td>
<td>1.000 -0.751E-05</td>
</tr>
<tr>
<td>Well I.D.</td>
<td>Time</td>
<td>Depth to Free Product (feet)</td>
<td>Depth to Water (feet)</td>
<td>Depth to Well Bottom (feet)</td>
</tr>
<tr>
<td>----------</td>
<td>------</td>
<td>------------------------------</td>
<td>----------------------</td>
<td>-----------------------------</td>
</tr>
<tr>
<td>MU-14</td>
<td>09:12</td>
<td>39.63</td>
<td>31.23</td>
<td></td>
</tr>
<tr>
<td>MU-13</td>
<td>09:40</td>
<td>31.26</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-6</td>
<td>10:43</td>
<td>31.26</td>
<td>20.09</td>
<td></td>
</tr>
<tr>
<td>MU-5</td>
<td>10:02</td>
<td>20.96</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-2</td>
<td>10:06</td>
<td>COVERED WITH ICE</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-4</td>
<td>10:09</td>
<td>30.19</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-11</td>
<td>16:14</td>
<td>28.51</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-16</td>
<td>10:18</td>
<td>29.71</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-15</td>
<td>10:23</td>
<td>30.93</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-9</td>
<td>10:26</td>
<td>28.65</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-10</td>
<td>10:37</td>
<td>27.14</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-12</td>
<td>10:32</td>
<td>27.53</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-3</td>
<td>10:35</td>
<td>27.62</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-8</td>
<td>10:43</td>
<td>28.45</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-7</td>
<td>10:48</td>
<td>27.05</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MU-1</td>
<td>11:05</td>
<td>27.62</td>
<td></td>
<td></td>
</tr>
<tr>
<td>FND</td>
<td>10:50</td>
<td>27.62</td>
<td></td>
<td></td>
</tr>
<tr>
<td>BCK</td>
<td>11:30</td>
<td>39.62</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Well I.D.</td>
<td>Time</td>
<td>Depth to Free Product (feet)</td>
<td>Depth to Water (feet)</td>
<td>Depth to Well Bottom (feet)</td>
</tr>
<tr>
<td>----------</td>
<td>----------</td>
<td>-----------------------------</td>
<td>----------------------</td>
<td>---------------------------</td>
</tr>
<tr>
<td>SVE 4</td>
<td>09:51</td>
<td>28.92</td>
<td>28.06</td>
<td>28.76</td>
</tr>
<tr>
<td>SVE 1</td>
<td>09:54</td>
<td>28.92</td>
<td>28.06</td>
<td></td>
</tr>
<tr>
<td>SVE 2</td>
<td>09:55</td>
<td>28.72</td>
<td>28.32</td>
<td></td>
</tr>
<tr>
<td>SVE 3</td>
<td>09:59</td>
<td>28.32</td>
<td>28.06</td>
<td></td>
</tr>
<tr>
<td>MIV 3</td>
<td>10:02</td>
<td>28.32</td>
<td>28.06</td>
<td></td>
</tr>
<tr>
<td>MIV 9</td>
<td>09:51</td>
<td>28.92</td>
<td>28.06</td>
<td></td>
</tr>
<tr>
<td>Project</td>
<td>Elks Court</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>---------</td>
<td>-------------</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Location</td>
<td>Anchor KS</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Client</td>
<td>Nustar</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Purpose of Observations</td>
<td>SUE Pilot Test</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ash Creek Representative</td>
<td>Paul Ware</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Contractor</td>
<td>Dave Cole</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Contractor Rep.</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Date</td>
<td>2013/12/13</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Our firm's professionals are represented on site solely to observe operations of the contractor identified, to form opinions about the adequacy of those operations, and to report those opinions to our client. The presence and activities of our field representatives do not relieve any contractor from its obligation to meet contractual requirements. The contractor retains sole responsibility for site safety and the methods, operations, and sequence of construction. Unless signed by the Ash Creek Associates Project Manager, this report is preliminary. A preliminary report is provided solely as evidence that field observation was performed. Observations and/or conclusions and/or recommendations conveyed in the final report may vary from and shall take precedence over those included in a preliminary report.

<table>
<thead>
<tr>
<th>Time</th>
<th>Event Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>09:00</td>
<td>Arrived, began set up, cleaned wells SE1 1.8&quot;, SE1 0.7&quot;</td>
</tr>
<tr>
<td>10:00</td>
<td>Set up Clayton water well, cleaned wells, set up flagging staff</td>
</tr>
<tr>
<td>10:02</td>
<td>Spoke to operator at 10:02, set up #1 sample, 5.5 cm (0.2&quot;)</td>
</tr>
<tr>
<td>10:09</td>
<td>P10 was 39.4, 1.0&quot;, VAC at wells 1.0&quot;, CSE1 0.7&quot;, MW 0.0&quot;</td>
</tr>
<tr>
<td>10:12</td>
<td>VAC at wells 0.7&quot;, CSE1 0.7&quot;, MW 0.0&quot;</td>
</tr>
<tr>
<td>10:14</td>
<td>VAC at wells 0.7&quot;, CSE1 0.7&quot;, MW 0.0&quot;</td>
</tr>
<tr>
<td>10:15</td>
<td>VAC at wells 0.7&quot;, CSE1 0.7&quot;, MW 0.0&quot;</td>
</tr>
<tr>
<td>10:20</td>
<td>VAC at wells 0.7&quot;, CSE1 0.7&quot;, MW 0.0&quot;</td>
</tr>
<tr>
<td>10:20</td>
<td>Increased VAC press. at 10:27, 11:11&quot;</td>
</tr>
</tbody>
</table>

**By**

[Signature]
APEX COMPANIES, LLC REPRESENTATIVE

**Reviewed By**

[Signature]
APEX COMPANIES, LLC PROJECT MANAGER
11:13  PID 312.7  Vac at wells  SVE1 1.4"  SVE3 0.3"  SVE4 0.8"  MW9 0.0"  MW3 0.0"

11:17  PID 191.4  Vac at wells  SVE1 1.3  SVE3 0.0  SVE4 0.8  MW9 0.0  MW3 0.0

12:13  PID 243.4  Vac at wells  SVE1 1.3  SVE3 0.0  SVE4 0.7  MW9 0.0  MW3 0.0

12:16  Vac at blower  7.1 cfm  115.0"  SVE1 1.3  SVE3 0.0  SVE4 0.7  MW9 0.0  MW3 0.0

12:17  Increased vac —  7.5 cfm  115.2"  at 12:15

12:20  PID 237.3  Vac at wells  SVE1 1.3  SVE3 0.0  SVE4 0.7  MW9 0.0  MW3 0.0

12:45  PID 282.2  Vac at blower  7.0 cfm  115.2"  SVE1 1.3  SVE3 0.0  SVE4 0.7  MW9 0.0  MW3 0.0

12:46  Lowed vac to 98"  at 12:45  7.0 cfm

12:48  PID 521.0  Vac at wells  SVE1 0.8  SVE3 0.0  SVE4 0.7  MW9 0.0  MW3 0.0

12:49  Raised vac to 95"  —

12:50  Vac at blower  7.0 cfm  Vac at wells  SVE1 0.9"  SVE4 0.2"  MW9 0.0  MW3 0.0

Raised vac loss to 115"  to take Field Sample
PID was 25.8  Oper. Value for 30 sec.  Sample Time 12:57

BY

ASH CREEK ASSOCIATES REPRESENTATIVE

REVIEWED BY

ASH CREEK ASSOCIATES PROJECT MANAGER
<table>
<thead>
<tr>
<th>Time</th>
<th>Event Description</th>
<th>Details</th>
</tr>
</thead>
<tbody>
<tr>
<td>14:15</td>
<td>Valve at Blower 7&quot;</td>
<td>1/16&quot;</td>
</tr>
<tr>
<td></td>
<td>34.8&quot;</td>
<td>Vacuum Wells</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 2 3 10&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 3 0.0&quot;</td>
</tr>
<tr>
<td>14:32</td>
<td>Valve at Blower 7&quot;</td>
<td>34.1&quot;</td>
</tr>
<tr>
<td></td>
<td>Vacuum Wells</td>
<td>Svc 1 2 1.2&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 2 0.8&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 3 0.0&quot;</td>
</tr>
<tr>
<td>14:45</td>
<td>Valve at Blower 7&quot;</td>
<td>35.2</td>
</tr>
<tr>
<td></td>
<td>Vacuum Wells</td>
<td>Svc 1 2.3&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 2 0.8&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 3 0.1&quot;</td>
</tr>
<tr>
<td></td>
<td>Increase Valve to 7.5&quot;</td>
<td></td>
</tr>
<tr>
<td>15:05</td>
<td>Valve at Blower 11&quot;</td>
<td>55.5&quot;</td>
</tr>
<tr>
<td></td>
<td>Vacuum Wells</td>
<td>Svc 1 5.4&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 2 5.0&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 3 0.3&quot;</td>
</tr>
<tr>
<td>15:15</td>
<td>Valve at Blower 11&quot;</td>
<td>55.4&quot;</td>
</tr>
<tr>
<td></td>
<td>Vacuum Wells</td>
<td>Svc 1 5.2&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 2 5.1&quot;</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Svc 3 0.3&quot;</td>
</tr>
<tr>
<td>15:20</td>
<td>Increase Valve to 15&quot;</td>
<td>75.1&quot;</td>
</tr>
</tbody>
</table>

By

[Signature]
APEX COMPANIES, LLC REPRESENTATIVE

Reviewed by

[Signature]
APEX COMPANIES, LLC PROJECT MANAGER
**Air - Chain of Custody Record & Analytical Service Request**

---

**Company Name & Address (Reporting Information)**
APEX Companies
305 SW First Ave
Portland OR 97201

**Project Manager**
Sam Jackson

**Phone** 503-924-4704 x111

**Email Address for Head: Reporting**
Jackson@APEXcos.com

---

**ALS Contact**

---

**Analysis Method**
EPA TO-15

---

**Sample(s) (Print & Sign)**
Paul Wood

---

### Client Sample ID

<table>
<thead>
<tr>
<th>Laboratory ID Number</th>
<th>Date Collected</th>
<th>Time Collected</th>
<th>Canister ID (Bar code # - NC, SC, etc.)</th>
<th>Flow Controller ID (Bar code # - FC #)</th>
<th>Canister Start Pressure (Hg)</th>
<th>Canister End Pressure (Hg)</th>
<th>Sample Volume</th>
</tr>
</thead>
<tbody>
<tr>
<td>JWE 2 Initial</td>
<td>11/25</td>
<td>10:29</td>
<td>15C000878</td>
<td>AVE028569</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>JWE 2 (compaction)</td>
<td>12/10-13</td>
<td>12:54</td>
<td>15C008861</td>
<td>AVE02461</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

---

**Report Tier Levels - please select**

<table>
<thead>
<tr>
<th>Tier I - Results (Default if not specified)</th>
<th>Tier II - Results + QC, Calibration, Summaries</th>
<th>Tier III - Results + QC &amp; Calibration Summaries</th>
<th>Tier IV - Data Validation Package 10% Surcharge</th>
</tr>
</thead>
</table>

---

**EDD required** YES / No

---

**Units:**

---

**Chain of Custody Seal:** (Circle)
- INTACT
- BROKEN
- ABSENT

---

**Project Requirements** (WFLs, QAPP)

---

**Relinquished by (Signature):**

---

**Date:** 12/10/13 **Time:** 17:00

---

**Received by:** (Signature)

---

**Date:** **Time:**

---

**Relinquished by (Signature):**

---

**Date:** **Time:**

---

**Date:** **Time:**

---

**Cooler / Blank Temperature:** °C
Appendix B

Analytical Laboratory Report – SVE Effluent Data
Appendix B — Laboratory Analytical Reports and Quality Assurance/Quality Control Results Summary

1.0 Introduction

This appendix documents the results of a quality assurance (QA) review of the analytical data for two air samples that were collected during the SVE Pilot Test performed December 9 through 11, 2013 at the Quail Crossing neighborhood (the Neighborhood) in Andover, Kansas (Site). The air samples were analyzed by ALS of Simi Valley, California.

The QA review outlines the applicable quality control criteria utilized during the data review process, as well as any deviations from those criteria. Examination and validation of the laboratory summary reports includes:

- Analytical methods;
- Reporting limits;
- Detection limits and estimated concentrations;
- Sample holding times;
- Custody records and sample receipt;
- Spikes, blanks, and surrogates; and
- Duplicates.

The QA review did not include a review of calibration or raw data. Section 2.0 lists the analytical methods used in sample analysis. Section 3.0 defines the QA terms used in this report. Section 4.0 provides the QA results for each sampling event.

2.0 Analytical Methods

Chemical analyses performed on air samples consisted of the following:

- 1,2,4-trimethylbenzene, 1,2-dichloroethane, 1,3,5-trimethylbenzene, BTEX, naphthalene, n-butylbenzene, and n-propylbenzene by EPA Method TO-15.

3.0 Quality Assurance Objectives and Review Procedures

The general QA objectives for this project were to develop and implement procedures for obtaining, evaluating, and confirming the usability of environmental data of a specified quality. To collect such information, analytical data must have an appropriate degree of accuracy and reproducibility, samples
collected must be representative of actual field conditions, and samples must be collected and analyzed using unbroken chain-of-custody (COC) procedures.

Reporting limits and analytical results were compared to action levels for each parameter in each media of concern. Precision, accuracy, representativeness, completeness, and comparability parameters used to indicate data quality are defined below.

**Reporting Limits.** Method reporting limits (MRLs) are set by the laboratory and are based on instrumentation abilities and sample matrix. In some cases, the MRLs are raised due to high concentrations of analytes in the samples or matrix interferences.

**Detection Limits and Estimated Concentrations.** The method detection limit (MDL) is the lowest quantity of a substance that can be distinguished from the absence of that substance within a stated confidence limit. The MDL is estimated from the mean of the blank, the standard deviation of the blank and some confidence factor.

**Holding Times.** Holding times are the length of time a sample can be stored after collection and prior to analysis without significantly affecting the analytical results.

**Custody Records and Sample Receipt.** COC refers to the document or paper trail showing the collection, custody, control, transfer, analysis, and disposition of physical materials. The sample receipt identifies the condition of samples upon arrival at the analytical laboratory.

**Method Blanks.** A method, or laboratory, blank is a sample prepared in the laboratory along with the field samples and analyzed for the same parameters at the same time. It is used to assess for laboratory introduced contamination.

**Laboratory Control Sample.** A laboratory control sample (LCS) is analyzed by the laboratory to assess the accuracy of the analytical equipment. The sample is prepared from an analyte-free matrix that is then spiked with known levels of the constituents of interest (i.e., a standard). The concentrations are measured and the results compared to the known spiked levels. This comparison is expressed as percent recovery.

**Laboratory Control Sample Duplicate.** A laboratory control sample duplicate (LCSD) is prepared and analyzed along with the LCS. The LCS and LCSD data are compared to assess the precision of the analytical method (i.e., the relative percent difference [RPD]).

**Matrix Spike Analyses.** Matrix spike (MS) analyses are performed on samples submitted to the laboratory that are of the same matrix as the field sample. The MS sample is spiked with known levels of the
Appendix B — Laboratory Analytical Reports and Quality Assurance/Quality Control Results Summary

constituents of interest and analyzed to assess the potential for matrix interference with recovery or detection of the constituents of interest and the accuracy of the determination. The spiked sample results are compared to the expected result (i.e., sample concentration plus spike amount) and reported as percent recovery.

**Lab Duplicate.** A laboratory duplicate is a second analysis of a QA/QC sample, which serves as an internal check on laboratory quality as well as potential variability of the sample matrix. The laboratory duplicate is analyzed and compared to the primary sample results to assess the precision of the analytical method. This comparison can be expressed by the RPD between the primary and duplicate samples.

**Surrogate Recovery.** Surrogates are organic compounds that are similar in chemical composition to the analytes of interest and spiked into environmental and batch QC samples prior to sample preparation and analysis. Surrogate recoveries for environmental samples are used to evaluate matrix interference on a sample-specific basis.

**Field Duplicate.** A field duplicate is a second field sample collected from a sampling location (e.g., a well or soil core). Field duplicate samples serve as a check on laboratory quality as well as potential variability of the sample matrix. The field duplicate is analyzed and compared to the primary sample to assess the precision of the analytical method. This comparison can be expressed by the RPD between the original and duplicate samples.

**Trip Blanks.** A trip blank is a sample prepared in the laboratory that is shipped along with the sample bottles to the field, kept with soil and groundwater samples during collection, and shipped back to the laboratory with the field samples. The trip blank is analyzed for constituents of interest, along with the primary samples, to assess if detected contaminants may have been the result of contamination of the samples during transport or storage.

**Equipment Blank.** An equipment blank is a sample collected in the field along with the primary samples and analyzed for the same parameters. Equipment blanks are collected by pouring deionized water over or through decontaminated equipment used to collect the samples, into laboratory supplied containers. The equipment blank is used to assess if field samples may have been affected by inadequate decontamination of field equipment.
Appendix B — Laboratory Analytical Reports and Quality Assurance/Quality Control Results Summary

4.0 QA/QC Review Results

The data reviewed includes two air samples collected on December 10, 2013. Samples were analyzed for 1,2,4-trimethylbenzene, 1,2-dichloroethane, 1,3,5-trimethylbenzene, BTEX, naphthalene, n-butylbenzene, and n-propylbenzene by EPA Method TO-15.

Reporting Limits. MRLs were not elevated for samples. No data are flagged.

Detection Limits and Estimated Concentrations. Due to the high concentrations of benzene and toluene in the samples, the detection limits were elevated for other VOCs; however, the detection limits were sufficient for the intended use.

Holding Times. The samples were analyzed within applicable holding times.

Custody Records and Sample Receipt. The samples were received at ambient temperature and consistent with the accompanying COC.

Method Blanks. No compounds were detected in the method blanks. No data are flagged.

Laboratory Control Sample. Percent recoveries of the LCS were within control limits.

Laboratory Control Sample Duplicate. Percent recoveries of the LCSD were within control limits. LCS/LCSD RPDs were within control limits for all tested compounds.

Lab Duplicate. No lab duplicate was analyzed.

Field Duplicate. Per KDHE guidance, no field duplicate was analyzed.

Trip Blank. No trip blank was analyzed.

Equipment Blank. No reusable equipment was utilized during this sampling event; therefore, an equipment blank was not collected.

Conclusion. In conclusion, the overall QA objectives have been met, and the data are of adequate quality for use in this project.
LABORATORY REPORT

December 27, 2013

Samuel Jackson
Apex Companies, LLC
3015 SW First Avenue
Portland, OR 97201-4707

RE: ANDOVER / COLT COURT / 320001641-04.002

Dear Samuel:

Enclosed are the results of the samples submitted to our laboratory on December 11, 2013. For your reference, these analyses have been assigned our service request number P1305477.

All analyses were performed according to our laboratory’s NELAP and DoD-ELAP-approved quality assurance program. The test results meet requirements of the current NELAP and DoD-ELAP standards, where applicable, and except as noted in the laboratory case narrative provided. For a specific list of NELAP and DoD-ELAP-accredited analytes, refer to the certifications section at www.alsglobal.com. Results are intended to be considered in their entirety and apply only to the samples analyzed and reported herein.

If you have any questions, please call me at (805) 526-7161.

Respectfully submitted,

ALS | Environmental

For Samantha Henningsen
Project Manager
CASE NARRATIVE

The samples were received intact under chain of custody on December 11, 2013 and were stored in accordance with the analytical method requirements. Please refer to the sample acceptance check form for additional information. The results reported herein are applicable only to the condition of the samples at the time of sample receipt.

Volatile Organic Compound Analysis

The samples were analyzed for selected volatile organic compounds in accordance with EPA Method TO-15 from the Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air, Second Edition (EPA/625/R-96/010b), January, 1999. This procedure is described in laboratory SOP VOA-TO15. The analytical system was comprised of a gas chromatograph / mass spectrometer (GC/MS) interfaced to a whole-air preconcentrator. This method is not included on the laboratory’s AIHA-LAP scope of accreditation. Any analytes flagged with an X are not included on the laboratory’s NELAP or DoD-ELAP scope of accreditation.

The Summa canisters were cleaned, prior to sampling, down to the method reporting limit (MRL) reported for this project. Please note, projects which require reporting below the MRL could have results between the MRL and method detection limit (MDL) that are biased high.

The results of analyses are given in the attached laboratory report. All results are intended to be considered in their entirety, and ALS Environmental (ALS) is not responsible for utilization of less than the complete report.

Use of ALS Environmental (ALS)’s Name. Client shall not use ALS’s name or trademark in any marketing or reporting materials, press releases or in any other manner (“Materials”) whatsoever and shall not attribute to ALS any test result, tolerance or specification derived from ALS’s data (“Attribution”) without ALS’s prior written consent, which may be withheld by ALS for any reason in its sole discretion. To request ALS’s consent, Client shall provide copies of the proposed Materials or Attribution and describe in writing Client’s proposed use of such Materials or Attribution. If ALS has not provided written approval of the Materials or Attribution within ten (10) days of receipt from Client, Client’s request to use ALS’s name or trademark in any Materials or Attribution shall be deemed denied. ALS may, in its discretion, reasonably charge Client for its time in reviewing Materials or Attribution requests. Client acknowledges and agrees that the unauthorized use of ALS’s name or trademark may cause ALS to incur irreparable harm for which the recovery of money damages will be inadequate. Accordingly, Client acknowledges and agrees that a violation shall justify preliminary injunctive relief. For questions contact the laboratory.
### Certifications, Accreditations, and Registrations

<table>
<thead>
<tr>
<th>Agency</th>
<th>Web Site</th>
<th>Number</th>
</tr>
</thead>
<tbody>
<tr>
<td>AIHA</td>
<td><a href="http://www.aihaaccreditedlabs.org">http://www.aihaaccreditedlabs.org</a></td>
<td>101661</td>
</tr>
<tr>
<td>Arizona DHS</td>
<td><a href="http://www.azdhs.gov/lab/license/env.htm">http://www.azdhs.gov/lab/license/env.htm</a></td>
<td>AZ0694</td>
</tr>
<tr>
<td>Florida DOH (NELAP)</td>
<td><a href="http://www.doh.state.fl.us/lab/EnvLabCert/WaterCert.htm">http://www.doh.state.fl.us/lab/EnvLabCert/WaterCert.htm</a></td>
<td>E871020</td>
</tr>
<tr>
<td>Minnesota DOH (NELAP)</td>
<td><a href="http://www.health.state.mn.us/accreditation">http://www.health.state.mn.us/accreditation</a></td>
<td>581572</td>
</tr>
<tr>
<td>New Jersey DEP (NELAP)</td>
<td><a href="http://www.nj.gov/dep/oqa/">http://www.nj.gov/dep/oqa/</a></td>
<td>CA009</td>
</tr>
<tr>
<td>New York DOH (NELAP)</td>
<td><a href="http://www.wadsworth.org/labcert/elap/elap.html">http://www.wadsworth.org/labcert/elap/elap.html</a></td>
<td>11221</td>
</tr>
<tr>
<td>Oregon PHD (NELAP)</td>
<td><a href="http://public.health.oregon.gov/LaboratoryServices/EnvironmentalLaboratoryAccreditation/Pages/index.aspx">http://public.health.oregon.gov/LaboratoryServices/EnvironmentalLaboratoryAccreditation/Pages/index.aspx</a></td>
<td>CA200007</td>
</tr>
<tr>
<td>Pennsylvania DEP</td>
<td><a href="http://www.depweb.state.pa.us/labs">http://www.depweb.state.pa.us/labs</a></td>
<td>68-03307</td>
</tr>
<tr>
<td></td>
<td>(Registration)</td>
<td></td>
</tr>
</tbody>
</table>

Analyses were performed according to our laboratory’s NELAP and DoD-ELAP approved quality assurance program. A complete listing of specific NELAP and DoD-ELAP certified analytes can be found in the certifications section at [www.alsglobal.com](http://www.alsglobal.com), or at the accreditation body’s website.

Each of the certifications listed above have an explicit Scope of Accreditation that applies to specific matrices/methods/analytes; therefore, please contact the laboratory for information corresponding to a particular certification.
# ALS ENVIRONMENTAL

## DETAIL SUMMARY REPORT

**Client:** Apex Companies, LLC  
**Project ID:** ANDOVER / COLT COURT / 320001641-04.002  
**Service Request:** P1305477

**Date Received:** 12/11/2013  
**Time Received:** 09:45

<table>
<thead>
<tr>
<th>Client Sample ID</th>
<th>Lab Code</th>
<th>Matrix</th>
<th>Date Collected</th>
<th>Time Collected</th>
<th>Container ID</th>
<th>P1 (psig)</th>
<th>P2 (psig)</th>
<th>Comment</th>
</tr>
</thead>
<tbody>
<tr>
<td>SVE 2 Initial</td>
<td>P1305477-001</td>
<td>Air</td>
<td>12/10/2013</td>
<td>10:29</td>
<td>1SC00878</td>
<td>13.22</td>
<td>13.22</td>
<td>X</td>
</tr>
<tr>
<td>SVE 2 Completion</td>
<td>P1305477-002</td>
<td>Air</td>
<td>12/10/2013</td>
<td>12:54</td>
<td>1SC00861</td>
<td>14.22</td>
<td>14.22</td>
<td>X</td>
</tr>
</tbody>
</table>
### Air - Chain of Custody Record & Analytical Service Request

**Company Name & Address (Reporting Information)**
APK Companies
2015 SW First Ave
Portland OR 97201

**Project Manager**
Sam Jackson

**Email Address for Result Reporting**
skw540@apkcx.com

**Client Sample ID**

<table>
<thead>
<tr>
<th>Sample ID</th>
<th>ID Number</th>
<th>Date Collected</th>
<th>Time Collected</th>
</tr>
</thead>
<tbody>
<tr>
<td>SVE 2 Initial</td>
<td>1</td>
<td>12-10-13</td>
<td>12:09</td>
</tr>
<tr>
<td>SVE 2 Control</td>
<td>2</td>
<td>12-10-13</td>
<td>12:54</td>
</tr>
</tbody>
</table>

**Sample Volume**

- [ ]

**Analysis Method**

**Prep**

**Sample (Print & Sign)**
Patti Ward

**Relinquished by (Signature)**

**Relinquished by (Signature)**

**Report Tier Levels - please select**

- Tier I - Results (Default if not specified)
- Tier II - Results + QC & Calibration Summaries
- Tier III - Results + QC & Calibration Summaries

**EDD required YES / No**

**Type:**

**Units:**

**Chain of Custody Seal:** (Circle)

- INTACT
- BROKEN
- ABSENT

**Project Requirements**

- MRLs, GAPP

**Status**

**Cooler/Blank Temperature**

<table>
<thead>
<tr>
<th>Date</th>
<th>Time</th>
</tr>
</thead>
<tbody>
<tr>
<td>12-10-13</td>
<td>17:00</td>
</tr>
</tbody>
</table>

---

**Comments**

- e.g. Actual Preservative or specific instructions
# ALS Environmental

## Sample Acceptance Check Form

**Client:** Apex Companies, LLC  
**Work order:** P1305477  
**Project:** ANDOVER / COLT COURT / 320001641-04.002  
**Sample(s) received on:** 12/11/13  
**Date opened:** 12/11/13  
**by:** RMARTENIES

---

**Note:** This form is used for all samples received by ALS. The use of this form for custody seals is strictly meant to indicate presence/absence and not as an indication of compliance or nonconformity. Thermal preservation and pH will only be evaluated either at the request of the client and/or as required by the method/SOP.

### 1. Were sample containers properly marked with client sample ID?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 2. Container(s) supplied by ALS?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 3. Did sample containers arrive in good condition?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 4. Were chain-of-custody papers used and filled out?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 5. Did sample container labels and/or tags agree with custody papers?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 6. Was sample volume received adequate for analysis?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 7. Are samples within specified holding times?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 8. Was proper temperature (thermal preservation) of cooler at receipt adhered to?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 9. Was a trip blank received?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 10. Were custody seals on outside of cooler/Box?

<table>
<thead>
<tr>
<th>Location of seal(s)?</th>
<th>Sealing Lid?</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
</tr>
</tbody>
</table>

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 11. Do containers have appropriate preservation, according to method/SOP or Client specified information?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 12. Badges:

<table>
<thead>
<tr>
<th>Description</th>
<th>Required pH *</th>
<th>Received pH</th>
<th>Adjusted pH</th>
<th>VOA Headspace (Presence/Absence)</th>
<th>Receipt / Preservation Comments</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.0 L Source Can</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1.0 L Source Can</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### 13. Are the badges properly capped and intact?

- [ ] Yes  
- [ ] No  
- [ ] N/A

### 14. Are dual bed badges separated and individually capped and intact?

- [ ] Yes  
- [ ] No  
- [ ] N/A

---

### Explain any discrepancies: (include lab sample ID numbers):

- [ ] RSK - MEEPP, HCL (pH<2); RSK - CO2, (pH 5-8); Sulfur (pH>4)
Client: Apex Companies, LLC

Client Sample ID: SVE 2 Initial

Client Project ID: ANDOVER / COLT COURT / 320001641-04.002

ALS Project ID: P1305477

ALS Sample ID: P1305477-001

Test Code: EPA TO-15

Instrument ID: Tekmar AUTOCAN/Agilent 5973inert/6890N/MS8

Analyst: Wida Ang

Sample Type: 1.0 L Summa Canister

Volume(s) Analyzed: 0.00020 Liter(s)

Initial Pressure (psig): 13.22

Final Pressure (psig): 13.22

Canister Dilution Factor: 1.00

<table>
<thead>
<tr>
<th>CAS #</th>
<th>Compound</th>
<th>Result µg/m³</th>
<th>MRL µg/m³</th>
<th>Result ppbV</th>
<th>MRL ppbV</th>
<th>Data Qualifier</th>
</tr>
</thead>
<tbody>
<tr>
<td>107-06-2</td>
<td>1,2-Dichloroethane</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>620</td>
<td></td>
</tr>
<tr>
<td>71-43-2</td>
<td>Benzene</td>
<td>11,000</td>
<td>2,500</td>
<td>3,400</td>
<td>780</td>
<td></td>
</tr>
<tr>
<td>108-88-3</td>
<td>Toluene</td>
<td>24,000</td>
<td>2,500</td>
<td>6,400</td>
<td>660</td>
<td></td>
</tr>
<tr>
<td>100-41-4</td>
<td>Ethylbenzene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>580</td>
<td></td>
</tr>
<tr>
<td>179601-23-1</td>
<td>m,p-Xylenes</td>
<td>ND</td>
<td>5,000</td>
<td>ND</td>
<td>1,200</td>
<td></td>
</tr>
<tr>
<td>95-47-6</td>
<td>o-Xylene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>580</td>
<td></td>
</tr>
<tr>
<td>103-65-1</td>
<td>n-Propylbenzene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>510</td>
<td></td>
</tr>
<tr>
<td>108-67-8</td>
<td>1,3,5-Trimethylbenzene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>510</td>
<td></td>
</tr>
<tr>
<td>95-63-6</td>
<td>1,2,4-Trimethylbenzene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>510</td>
<td></td>
</tr>
<tr>
<td>91-20-3</td>
<td>Naphthalene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>480</td>
<td></td>
</tr>
<tr>
<td>104-51-8</td>
<td>n-Butylbenzene</td>
<td>ND</td>
<td>2,500</td>
<td>ND</td>
<td>460</td>
<td></td>
</tr>
</tbody>
</table>

ND = Compound was analyzed for, but not detected above the laboratory reporting limit.

MRL = Method Reporting Limit - The minimum quantity of a target analyte that can be confidently determined by the referenced method.
### RESULTS OF ANALYSIS

**Client:** Apex Companies, LLC  
**Client Sample ID:** SVE 2 Completion  
**Client Project ID:** ANDOVER / COLT COURT / 320001641-04.002  
**ALS Project ID:** P1305477  
**ALS Sample ID:** P1305477-002  

**Test Code:** EPA TO-15  
**Instrument ID:** Tekmar AUTOCAN/Agilent 5973inert/6890N/MS8  
**Analyst:** Wida Ang  
**Sample Type:** 1.0 L Summa Canister  
**Volume(s) Analyzed:** 0.00060 Liter(s)  

**Canister Dilution Factor:** 1.00

<table>
<thead>
<tr>
<th>CAS #</th>
<th>Compound</th>
<th>Result µg/m³</th>
<th>MRL µg/m³</th>
<th>Result ppbV</th>
<th>MRL ppbV</th>
<th>Data Qualifier</th>
</tr>
</thead>
<tbody>
<tr>
<td>107-06-2</td>
<td>1,2-Dichloroethane</td>
<td>ND</td>
<td>830</td>
<td>ND</td>
<td>210</td>
<td></td>
</tr>
<tr>
<td>71-43-2</td>
<td>Benzene</td>
<td>3,200</td>
<td>830</td>
<td>990</td>
<td>260</td>
<td></td>
</tr>
<tr>
<td>108-88-3</td>
<td>Toluene</td>
<td>18,000</td>
<td>830</td>
<td>4,900</td>
<td>220</td>
<td></td>
</tr>
<tr>
<td>100-41-4</td>
<td>Ethylbenzene</td>
<td>5,800</td>
<td>830</td>
<td>1,300</td>
<td>190</td>
<td></td>
</tr>
<tr>
<td>179601-23-1</td>
<td>m,p-Xylenes</td>
<td>13,000</td>
<td>1,700</td>
<td>3,000</td>
<td>380</td>
<td></td>
</tr>
<tr>
<td>95-47-6</td>
<td>o-Xylene</td>
<td>3,700</td>
<td>830</td>
<td>850</td>
<td>190</td>
<td></td>
</tr>
<tr>
<td>103-65-1</td>
<td>n-Propylbenzene</td>
<td>ND</td>
<td>830</td>
<td>ND</td>
<td>170</td>
<td></td>
</tr>
<tr>
<td>108-67-8</td>
<td>1,3,5-Trimethylbenzene</td>
<td>ND</td>
<td>830</td>
<td>ND</td>
<td>170</td>
<td></td>
</tr>
<tr>
<td>95-63-6</td>
<td>1,2,4-Trimethylbenzene</td>
<td>1,000</td>
<td>830</td>
<td>200</td>
<td>170</td>
<td></td>
</tr>
<tr>
<td>91-20-3</td>
<td>Naphthalene</td>
<td>ND</td>
<td>830</td>
<td>ND</td>
<td>160</td>
<td></td>
</tr>
<tr>
<td>104-51-8</td>
<td>n-Butylbenzene</td>
<td>ND</td>
<td>830</td>
<td>ND</td>
<td>150</td>
<td></td>
</tr>
</tbody>
</table>

ND = Compound was analyzed for, but not detected above the laboratory reporting limit.  
MRL = Method Reporting Limit - The minimum quantity of a target analyte that can be confidently determined by the referenced method.
## RESULTS OF ANALYSIS

**Client:** Apex Companies, LLC  
**Client Sample ID:** Method Blank  
**Client Project ID:** ANDOVER / COLT COURT / 320001641-04.002  
**ALS Project ID:** P1305477

**Instrument ID:** Tekmar AUTOCAN/Agilent 5973inert/6890N/MS8  
**ALS Sample ID:** P131219-MB  
**Date Collected:** NA  
**Date Received:** NA  
**Date Analyzed:** 12/19/13

### Test Notes:
- **Canister Dilution Factor:** 1.00
- **CAS #**  
- **Compound**  
- **Result µg/m³**  
- **MRL µg/m³**  
- **Result ppbV**  
- **MRL ppbV**  
- **Data Qualifier**

<table>
<thead>
<tr>
<th>CAS #</th>
<th>Compound</th>
<th>Result µg/m³</th>
<th>MRL µg/m³</th>
<th>Result ppbV</th>
<th>MRL ppbV</th>
<th>Data Qualifier</th>
</tr>
</thead>
<tbody>
<tr>
<td>107-06-2</td>
<td>1,2-Dichloroethane</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>71-43-2</td>
<td>Benzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.16</td>
<td></td>
</tr>
<tr>
<td>108-88-3</td>
<td>Toluene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.13</td>
<td></td>
</tr>
<tr>
<td>100-41-4</td>
<td>Ethylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>179601-23-1</td>
<td>m,p-Xylenes</td>
<td>ND</td>
<td>1.0</td>
<td>ND</td>
<td>0.23</td>
<td></td>
</tr>
<tr>
<td>95-47-6</td>
<td>o-Xylene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>103-65-1</td>
<td>n-Propylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>108-67-8</td>
<td>1,3,5-Trimethylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>95-63-6</td>
<td>1,2,4-Trimethylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>91-20-3</td>
<td>Naphthalene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.095</td>
<td></td>
</tr>
<tr>
<td>104-51-8</td>
<td>n-Butylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.091</td>
<td></td>
</tr>
</tbody>
</table>

ND = Compound was analyzed for, but not detected above the laboratory reporting limit.  
MRL = Method Reporting Limit - The minimum quantity of a target analyte that can be confidently determined by the referenced method.
### RESULTS OF ANALYSIS

**Client:** Apex Companies, LLC  
**Client Sample ID:** Method Blank  
**Client Project ID:** ANDOVER / COLT COURT / 320001641-04.002  
**ALS Project ID:** P1305477  
**ALS Sample ID:** P131226-MB

**Test Code:** EPA TO-15  
**Instrument ID:** Tekmar AUTOCAN/Agilent 5973inert/6890N/MS8  
**Test Notes:**

<table>
<thead>
<tr>
<th>CAS #</th>
<th>Compound</th>
<th>Result $\mu g/m^3$</th>
<th>MRL $\mu g/m^3$</th>
<th>Result ppbV</th>
<th>MRL ppbV</th>
<th>Data Qualifier</th>
</tr>
</thead>
<tbody>
<tr>
<td>107-06-2</td>
<td>1,2-Dichloroethane</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>71-43-2</td>
<td>Benzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.16</td>
<td></td>
</tr>
<tr>
<td>108-88-3</td>
<td>Toluene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.13</td>
<td></td>
</tr>
<tr>
<td>100-41-4</td>
<td>Ethylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>179601-23-1</td>
<td>m,p-Xylenes</td>
<td>ND</td>
<td>1.0</td>
<td>ND</td>
<td>0.23</td>
<td></td>
</tr>
<tr>
<td>95-47-6</td>
<td>o-Xylene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.12</td>
<td></td>
</tr>
<tr>
<td>103-65-1</td>
<td>n-Propylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>108-67-8</td>
<td>1,3,5-Trimethylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>95-63-6</td>
<td>1,2,4-Trimethylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>91-20-3</td>
<td>Naphthalene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.095</td>
<td></td>
</tr>
<tr>
<td>104-51-8</td>
<td>n-Butylbenzene</td>
<td>ND</td>
<td>0.50</td>
<td>ND</td>
<td>0.091</td>
<td></td>
</tr>
</tbody>
</table>

**Volume(s) Analyzed:** 1.00 Liter(s)  
**Date Analyzed:** 12/26/13  
**Date Received:** NA  
**Date Collected:** NA  
**Canister Dilution Factor:** 1.00

**Test Notes:**

ND = Compound was analyzed for, but not detected above the laboratory reporting limit.  
MRL = Method Reporting Limit - The minimum quantity of a target analyte that can be confidently determined by the referenced method.
# ALS ENVIRONMENTAL

## SURROGATE SPIKE RECOVERY RESULTS

Page 1 of 1

**Client:** Apex Companies, LLC  
**Client Project ID:** ANDOVER / COLT COURT / 320001641-04.002  
**ALS Project ID:** P1305477

Test Code: EPA TO-15  
Instrument ID: Tekmar AUTOCAN/Agilent 5973inert/6890N/MS8  
**Date(s) Collected:** 12/10/13  
**Date(s) Received:** 12/11/13  
**Sample Type:** 1.0 L Summa Canister(s)  
**Date(s) Analyzed:** 12/19 - 12/26/13

**Analyst:** Wida Ang  
**Test Notes:**

<table>
<thead>
<tr>
<th>Client Sample ID</th>
<th>ALS Sample ID</th>
<th>1,2-Dichloroethane-d4</th>
<th>Toluene-d8</th>
<th>Bromofluorobenzene</th>
</tr>
</thead>
<tbody>
<tr>
<td>Method Blank</td>
<td>P131219-MB</td>
<td>110</td>
<td>97</td>
<td>102</td>
</tr>
<tr>
<td>Method Blank</td>
<td>P131226-MB</td>
<td>115</td>
<td>95</td>
<td>105</td>
</tr>
<tr>
<td>Lab Control Sample</td>
<td>P131219-LCS</td>
<td>107</td>
<td>97</td>
<td>101</td>
</tr>
<tr>
<td>Lab Control Sample</td>
<td>P131226-LCS</td>
<td>111</td>
<td>87</td>
<td>104</td>
</tr>
<tr>
<td>SVE 2 Initial</td>
<td>P1305477-001</td>
<td>112</td>
<td>99</td>
<td>102</td>
</tr>
<tr>
<td>SVE 2 Completion</td>
<td>P1305477-002</td>
<td>118</td>
<td>100</td>
<td>102</td>
</tr>
</tbody>
</table>

Surrogate percent recovery is verified and accepted based on the on-column result.  
Reported results are shown in concentration units and as a result of the calculation, may vary slightly from the on-column percent recovery.
ALS ENVIRONMENTAL

LABORATORY CONTROL SAMPLE SUMMARY

Page 1 of 1

Client: Apex Companies, LLC

Client Sample ID: Lab Control Sample

Client Project ID: ANDOVER / COLT COURT / 320001641-04.002

ALS Project ID: P1305477

ALS Sample ID: P131219-LCS

Test Code: EPA TO-15

Instrument ID: Tekmar AUTOCAN/Agilent 5973inert/6890N/MS8

Analyst: Wida Ang

Sample Type: 1.0 L Summa Canister

Date Collected: NA

Date Received: NA

Date Analyzed: 12/19/13

Volume(s) Analyzed: 0.125 Liter(s)

<table>
<thead>
<tr>
<th>CAS #</th>
<th>Compound</th>
<th>Spike Amount µg/m³</th>
<th>Result µg/m³</th>
<th>% Recovery</th>
<th>ALS Acceptance Limits</th>
</tr>
</thead>
<tbody>
<tr>
<td>107-06-2</td>
<td>1,2-Dichloroethane</td>
<td>210</td>
<td>206</td>
<td>98</td>
<td>69-118</td>
</tr>
<tr>
<td>71-43-2</td>
<td>Benzene</td>
<td>220</td>
<td>200</td>
<td>91</td>
<td>69-117</td>
</tr>
<tr>
<td>108-88-3</td>
<td>Toluene</td>
<td>210</td>
<td>185</td>
<td>88</td>
<td>65-116</td>
</tr>
<tr>
<td>100-41-4</td>
<td>Ethylbenzene</td>
<td>212</td>
<td>197</td>
<td>93</td>
<td>66-119</td>
</tr>
<tr>
<td>179601-23-1</td>
<td>m,p-Xylenes</td>
<td>420</td>
<td>405</td>
<td>96</td>
<td>64-118</td>
</tr>
<tr>
<td>95-47-6</td>
<td>o-Xylene</td>
<td>206</td>
<td>199</td>
<td>97</td>
<td>65-120</td>
</tr>
<tr>
<td>103-65-1</td>
<td>n-Propylbenzene</td>
<td>202</td>
<td>194</td>
<td>96</td>
<td>65-121</td>
</tr>
<tr>
<td>108-67-8</td>
<td>1,3,5-Trimethylbenzene</td>
<td>212</td>
<td>206</td>
<td>97</td>
<td>64-125</td>
</tr>
<tr>
<td>95-63-6</td>
<td>1,2,4-Trimethylbenzene</td>
<td>212</td>
<td>213</td>
<td>100</td>
<td>64-131</td>
</tr>
<tr>
<td>91-20-3</td>
<td>Naphthalene</td>
<td>204</td>
<td>212</td>
<td>104</td>
<td>56-143</td>
</tr>
<tr>
<td>104-51-8</td>
<td>n-Butylbenzene</td>
<td>220</td>
<td>215</td>
<td>98</td>
<td>66-127</td>
</tr>
</tbody>
</table>

Laboratory Control Sample percent recovery is verified and accepted based on the on-column result.

Reported results are shown in concentration units and as a result of the calculation, may vary slightly.
### ALS ENVIRONMENTAL

**LABORATORY CONTROL SAMPLE SUMMARY**

---

**Client:** Apex Companies, LLC  
**Client Sample ID:** Lab Control Sample  
**Client Project ID:** ANDOVER / COLT COURT / 320001641-04.002

**ALS Sample ID:** P131226-LCS  
**ALS Project ID:** P1305477

| Test Code: | EPA TO-15 |  
| Instrument ID: | Tekmar AUTOCAN/Agilent 5973 inert/6890N/MS8 |  
| Analyst: | Wida Ang |  
| Date Received: | NA |  
| Sample Type: | 1.0 L Summa Canister |  
| Date Collected: | NA |  
| Date Analyzed: | 12/26/13 |  
| Volume(s) Analyzed: | 0.125 Liter(s) |

---

**Compound**  
Spike Amount | Result | % Recovery | ALS Acceptance Limits | Data Qualifier  
--- | --- | --- | --- | ---  
1,2-Dichloroethane | 210 µg/m³ | 198 µg/m³ | 94 | 69-118 |  
Benzene | 220 µg/m³ | 187 µg/m³ | 85 | 69-117 |  
Toluene | 210 µg/m³ | 154 µg/m³ | 73 | 65-116 |  
Ethylbenzene | 212 µg/m³ | 175 µg/m³ | 83 | 66-119 |  
m,p-Xylenes | 420 µg/m³ | 369 µg/m³ | 88 | 64-118 |  
o-Xylene | 206 µg/m³ | 180 µg/m³ | 87 | 65-120 |  
n-Propylbenzene | 202 µg/m³ | 174 µg/m³ | 86 | 65-121 |  
1,3,5-Trimethylbenzene | 212 µg/m³ | 184 µg/m³ | 87 | 64-125 |  
1,2,4-Trimethylbenzene | 212 µg/m³ | 193 µg/m³ | 91 | 64-131 |  
Naphthalene | 204 µg/m³ | 190 µg/m³ | 93 | 56-143 |  
n-Butylbenzene | 220 µg/m³ | 193 µg/m³ | 88 | 66-127 |  

Laboratory Control Sample percent recovery is verified and accepted based on the on-column result. Reported results are shown in concentration units and as a result of the calculation, may vary slightly.
Appendix C

Sampling and Analysis Plan
Appendix C – Sampling and Analysis Plan: Soil Investigation

1.0 Introduction

This sampling and analysis plan (SAP) presents the field and sampling procedures and the analytical testing program that will be used to perform the soil investigation in the Quail Crossing Neighborhood near Andover, Kansas (the Site). Quality assurance and quality control (QA/QC) procedures are also discussed in this appendix.

2.0 Field and Sampling Procedures

The scope of work includes completion of subsurface explorations and soil sampling. Data from these activities will be used to refine the understanding of the vertical and lateral extent of soil with gasoline constituents at the Site. These data will be used to evaluate cleanup options. The field and sampling procedures include:

- Advancement of borings using hollow-stem auger equipment;
- Collection of soil samples;
- Field screening of soil samples;
- Selection and containerizing soil samples for laboratory analysis;
- Sample management (e.g., containers, storage, and shipment);
- Sample location control;
- Boring abandonment;
- Decontamination procedures; and
- Handling of investigation-derived waste (IDW).

2.1 Preparatory Activities

Property Access. The soil investigation will be performed within: (1) the NuStar Pipeline right of way (ROW); and (2) the City of Andover ROW. NuStar executed an access agreement with the City of Andover in 2013.

Property Owner Notification. The owners of properties in the Quail Crossing Neighborhood will be notified of planned drilling activities a minimum of one week in advance of field work.
Appendix C – Sampling and Analysis Plan: Soil Investigation

Underground Utility Location. An underground utility locate request will be submitted through Kansas One-Call Service. A private underground utility locate will also be conducted prior to performing the subsurface work.

Health and Safety Plan. A Health and Safety Plan (HASP) has been prepared and is included as an Attachment A to this Appendix. A copy of the HASP will be maintained on site during field activities.

2.2 Soil Borings, Field Screening, and Soil Sample Collection

The soil borings will be advanced using a hollow-stem auger. Discrete soil samples will be collected, using a stainless steel split-spoon sampler at approximately 2.5-foot intervals, for lithologic description. The soil sampling procedure includes driving the split-spoon sampler into the soil a distance of 18 inches using a 140-pound hammer with a free-fall height of 30 inches (per ASTM D-1587 guidelines). Each soil sample will be logged in general accordance with ASTM method D2487/2488. Soil sample descriptions will include visual indications of petroleum impacts, if any. The soil samples will be field screened for volatile organic compounds (VOCs) and SPH using a photoionization detector (PID) and sheen tests, in accordance with Standard Operating Procedure (SOP) 2.1, included in this appendix. Based on field indications of petroleum impacts, soil samples will be submitted for chemical analysis of gasoline-range organics (GRO) and VOCs by Iowa Method OA-1 and Environmental Protection Agency (EPA) Method 8260B, respectively. If indications of petroleum impacts are observed (e.g., sheen, staining, or headspace vapors), additional soil samples will be collected from the interval that exhibits the maximum impact and from apparently non-impacted soil overlying and underlying the impacted interval. A field log of lithology and soil conditions will be maintained for each boring. Drilling equipment (i.e., auger stems, split-spoon sampler, etc.) will be decontaminated before and after each boring is advanced.

Soil samples intended for analysis of volatile organics will be collected using EPA Method 5035, as described in SOP 2.7, included in this attachment.

Lithology will be noted on the field logs and described in accordance with the Unified Soil Classification System (USCS). Features such as sorting, sedimentary features, mineralogy, degree of weathering, and contacts with other soil types will also be noted if relevant.

2.3 Exploration Abandonment and Sample Location Control.

Explorations will be abandoned by filling each exploration with grout or bentonite to approximately three feet bgs. The upper three feet will be backfilled to match surrounding grade and surface completion. Following abandonment, each location will be recorded using a handheld global positioning system (GPS) instrument.
2.4 Sample Management

**Soil Containers.** Clean sample containers will be provided by the analytical laboratory ready for sample collection, including preservative if required (the container requirements are listed in Table C-1). Specific container requirements for samples that will undergo multiple analyses will be discussed with the analytical laboratory prior to sample collection.

**Labeling Requirements.** A sample label will be affixed to each sample container before sample collection. Containers will be marked with the project number, sample number, date of collection, and the sampler’s initials.

**Sample Storage and Shipment.** Soil samples will be stored in a cooler chilled with ice or blue ice to 4 degrees Celsius (°C). The cooler lid will be sealed with chain-of-custody seals. The samples will be sent via overnight courier to the analytical laboratory for chemical analysis. Chain of custody will be maintained and documented at all times.

2.5 Decontamination Procedures

**Personnel Decontamination.** Personnel decontamination procedures depend on the level of protection specified for a given activity. The HASP will identify the appropriate level of protection for the type of work and expected field conditions involved in this project. In general, clothing and other protective equipment can be removed from the investigation area. Field personnel should thoroughly wash their hands and faces at the end of each day and before taking any work breaks.

**Sampling Equipment Decontamination.** To minimize cross-contamination between sampling events, cleaning of non-disposable items will consist of washing in a detergent (Alconox®) solution, rinsing with tap water, followed by a de-ionized (DI) water rinse. Decontamination water will be handled as described in Section 2.6.

**Drilling Equipment and Materials.** Decontamination procedures are designed to remove trace-level contaminants from drilling equipment to prevent the cross-contamination of exploration locations and samples. Drilling equipment shall be decontaminated using high-pressure washing, steam cleaning, or cleaning with detergent before and after use.

2.6 Handling of Investigation-Derived Waste

IDW will consist of soil from the soil cores and soil cuttings, decontamination water, and purge water. IDW will be placed in Department of Transportation (DOT)-approved containers. Each container will be labeled...
Appendix C – Sampling and Analysis Plan: Soil Investigation

with the project name, general contents, and date. The IDW will be stored at the Site pending proper disposal.

Disposable items, such as sample tubing, disposable bailers, bailer line, gloves, protective overalls (e.g., Tyvek®), paper towels, etc., will be placed in plastic bags after use and deposited in trash receptacles for disposal.

3.0 Analytical Testing Program

An analytical testing program will be performed to assess the chemical quality of soil samples collected as part of this project. Analytical laboratory QA/QC procedures are discussed in Section 4 of this appendix.

Table C-2 lists the proposed analytical methods and laboratory detection limit goals. Samples will be collected and handled using methods described in Section 2 of this appendix. Specific container and storage requirements for samples are listed in Table C-1. The samples will be analyzed for GRO by Iowa Method OA-1 and VOCs by EPA Method 8260B.

4.0 Quality Assurance Program

4.1 Quality Assurance Objectives for Data Management

The general QA objectives for this project are to develop and implement procedures for evaluating soil analytical data. To collect such information, analytical data must have an appropriate degree of accuracy and reproducibility, samples must be representative of actual field conditions, and samples must be collected and analyzed using unbroken chain-of-custody procedures (see Section 4.3).

Method detection limits (MDLs) and analytical results will be compared to action levels for each parameter in media of concern. The detection limits listed in Table C-2 are the expected detection limits, based upon laboratory calculations and experience.

Specific QA objectives are as follows:

1) Establish sampling techniques that will produce analytical data representative of the media being measured.

2) Collect and analyze a sufficient number of duplicate field samples to establish sampling precision. Laboratory duplicates of the same sample will provide a measure of precision within that sample (sample homogeneity).
3) Analyze a sufficient number of analytical duplicate samples to assess the performance of the analytical laboratory.

4) Analyze a sufficient number of duplicate, spiked, and check samples within the laboratory to evaluate results against numerical QA goals established for precision and accuracy.

Precision, accuracy, representativeness, completeness, and comparability parameters used to indicate data quality are defined below.

### 4.1.1 Precision

Precision is a measure of the reproducibility of data under a given set of conditions. Specifically, it is a quantitative measure of the variability of a group of measurements compared to their average value. For duplicate measurements, precision can be expressed as the relative percent difference (RPD). Analysis of field duplicate samples measures the precision of sampling.

### 4.1.2 Accuracy

Accuracy is the measure of error between the reported test results and the true sample concentration. True sample concentration is never known due to analytical limitations and error. Consequently, accuracy is inferred from the recovery data from spiked samples.

Because of difficulties with spiking samples in the field, the laboratory will spike samples. For analyses of less than five samples, matrix spikes (MS) may be performed on a batch basis.

Perfect accuracy is 100 percent recovery.

### 4.1.3 Representativeness

Representativeness is a measure of how closely the results reflect the actual concentration of the chemical parameters in the medium sampled. Sampling procedures as well as sample-handling protocols for storage, preservation, and transportation are designed to preserve the representativeness of the samples collected. Proper documentation will confirm that protocols are followed. This helps to assure sample identification and integrity.

Laboratory method blanks will be run in accordance with established laboratory protocols to ensure samples are not contaminated during sample preparation in the laboratory.
Appendix C – Sampling and Analysis Plan: Soil Investigation

4.1.4 Completeness

Completeness is defined as the percentage of measurements made which are judged to be valid. The completeness goal is essentially that a sufficient amount of valid data be generated to meet the closure requirements.

4.1.5 Comparability

Comparability is a qualitative parameter expressing the confidence with which one data set can be compared with another. The objective of this QA program is to assure that data developed during the investigation are comparable. Comparability of the data will be assured by using EPA-defined procedures which specify sample collection, handling, and analytical methods.

4.1.6 Documentation

Essentially, EPA Level III documentation will be generated during this investigation. This level of documentation is generally considered legally defensible and consists of the following:

- Holding times;
- Field duplicate data;
- Laboratory method blank data;
- Sample data;
- Matrix/surrogate spike data; and
- Duplicate sample data.

4.2 Sampling Procedures

Sampling procedures for soil are presented above in Section 2 of this appendix. These procedures are designed to ensure:

- Samples collected at the Site are consistent with project objectives; and
- Samples are identified, handled, and transported in a manner that does not alter the representativeness of the data from the actual Site conditions.

QA objectives for sample collection will be accomplished by evaluating the following items:

- **Duplicate Samples.** Duplicates will be submitted to evaluate the precision of soil analysis. Three field duplicates are included in the scope for this work.
Appendix C – Sampling and Analysis Plan: Soil Investigation

- Laboratory QA. Samples will be analyzed by ALS Laboratories, a KDHE-certified laboratory. The ALS Quality Assurance Program Plan has been provided to KDHE previously.

4.3 Sample and Document Custody Procedures

The various methods used to document field sample collection and laboratory operation are presented below.

4.3.1 Field Chain-of-Custody Procedures

Sample chain of custody refers to the process of tracking the possession of a sample from the time it is collected in the field through the laboratory analysis. A sample is considered to be under a person's custody if it is:

- In a person's physical possession;
- In view of the person after possession has been taken; or
- Secured by that person so no one can tamper with the sample, or secured by that person in an area restricted to authorized personnel.

A chain-of-custody form is used to record possession of a sample and to document analyses requested. Each time the sample bottles or samples are transferred between individuals, both the sender and receiver sign and date the chain-of-custody form. When a sample shipment is transported to the laboratory, a copy of the chain-of-custody form is included in the transport container (e.g., ice chest).

The chain-of-custody forms are used to record the following information:

- Sample identification number;
- Sample collector's signature;
- Date and time of collection;
- Description of sample;
- Analyses requested;
- Shipper's name and address;
- Receiver's name and address; and
- Signatures of persons involved in chain of custody.
4.3.2 Laboratory Operations

The analytical laboratory has a system in place for documenting the following laboratory information:

- Calibration procedures;
- Analytical procedures;
- Computational procedures;
- QC procedures;
- Bench data;
- Operating procedures or any changes to these procedures; and
- Laboratory notebook policy.

Laboratory chain-of-custody procedures provide the following:

- Identification of the responsible party (sample custodian) authorized to sign for incoming field samples and a log consisting of sequential lab tracking numbers; and
- Specification of laboratory sample custody procedures for sample handling, storage, and internal distribution for analysis.

4.3.3 Corrections to Documentation

Original data are recorded in field notes and on chain-of-custody forms using indelible ink. Documents will be retained even if they are illegible or contain inaccuracies that require correction.

If an error is made on a document, the individual making the entry will correct the document by crossing a line through the error, entering the correct information, and initialing and dating the correction. Any subsequent error discovered on a document is corrected, initialed, and dated by the person who made the entry.

4.4 Equipment Calibration Procedures and Frequency

Instruments and equipment used during this project will be operated, calibrated, and maintained according to the manufacturer's guidelines and recommendations. Operation, calibration, and maintenance will be performed by laboratory personnel fully trained in these procedures.
Appendix C – Sampling and Analysis Plan: Soil Investigation

4.5 Analytical Procedures

Samples will be analyzed using GRO by Iowa Method OA-1 and VOCs by EPA Method 8260B. Table C-2 lists analytical parameters and test methods.

4.6 Data Reduction, Validation, and Reporting

The Project Manager will assure validation of the analytical data. The laboratory generating analytical data for this project will be required to submit results that are supported by sufficient backup and QA/QC data to enable the reviewer to determine the quality of the data. Validity of the laboratory data will be determined based on the objectives outlined in Section 4.1 (above). Data validity will also be determined based upon the sampling procedures and documentation outlined in Sections 4.2 and 4.3 of this SAP. Upon completion of the review, the Project Manager will be responsible for assuring development of a QA/QC report on the analytical data. Data will be stored and maintained according to the standard procedures of the laboratory. The method of data reduction will be described in the final report.

4.7 Performance Audits

Performance audits are an integral part of an analytical laboratory’s SOPs and are available upon request.

4.8 Corrective Actions

If the QC audit detects unacceptable conditions or data, the Project Manager will be responsible for developing and initiating corrective action. The Project Manager will be notified if the nonconformance is significant or requires special expertise. Corrective action may include the following:

- Reanalyzing the samples, if holding time criteria permit;
- Resampling and analyzing;
- Evaluating and amending sampling and analytical procedures; and
- Accepting data and acknowledging level of uncertainty or inaccuracy by flagging the data.

4.9 Quality Assurance Reports

The Project Manager will prepare a QA/QC evaluation of the data collected during the investigation field activities for inclusion in the final report. In addition to an opinion regarding the validity of the data, the QA/QC evaluation will address the following:

- Adverse conditions or deviations from this SAP;
- Assessment of analytical data for precision, accuracy, and completeness;
Appendix C – Sampling and Analysis Plan: Soil Investigation

- Significant QA problems and recommended solutions; and
- Corrective actions taken for any problems previously identified.
### Table C-1
Analytical Methods - Sample Container Requirements

**NuStar Pipeline Operating Partnership L.P. - Quail Crossing Neighborhood**

**Andover, Kansas**

<table>
<thead>
<tr>
<th>Analysis</th>
<th>Method</th>
<th>Container</th>
<th>Preservative</th>
<th>Storage Temperature</th>
<th>Holding Time</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>VOCs</strong></td>
<td>EPA 8260B/5035</td>
<td>3 x VOAs, 1 x 4 oz jar</td>
<td>None</td>
<td>4°C</td>
<td>14 days</td>
</tr>
<tr>
<td><strong>TPH-GRO</strong></td>
<td>Iowa-OA1</td>
<td>1 x 4 oz jar</td>
<td>None</td>
<td>4°C</td>
<td>14 days</td>
</tr>
</tbody>
</table>

**Notes:**

1. EPA = U.S. Environmental Protection Agency.
2. TPH-GRO = Gasoline-range petroleum hydrocarbons.
3. VOCs = Volatile organic compounds.
4. °C = Degrees Celsius.

---

SVE Evaluation and Proposed Soil Investigation Report
Quail Crossing Neighborhood – Andover, Kansas
1641-04
Page 1 of 1
<table>
<thead>
<tr>
<th>Analyte</th>
<th>Method</th>
<th>Anticipated Number of Samples</th>
<th>Units</th>
<th>Detection Limit Goal</th>
</tr>
</thead>
<tbody>
<tr>
<td>Total Petroleum Hydrocarbons (TPH) Soil</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total Petroleum Hydrocarbons as Gasoline (TPH-GRO)</td>
<td>Iowa Method OA-1</td>
<td>50</td>
<td>mg/kg</td>
<td>0.5</td>
</tr>
<tr>
<td>Volatile Organic Compounds (VOCs) Soil</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Benzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>Cyclohexane</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>Ethylbenzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>Isopropylbenzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>Methylcyclohexane</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>Toluene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>Xylenes, Total</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.003</td>
</tr>
<tr>
<td>Naphthalene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>n-butylbenzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>1,2-Dichloroethane</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>1,3,5-trimethylbenzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>1,2,4-trimethylbenzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>2-butanone</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
<tr>
<td>n-Propylbenzene</td>
<td>EPA Method 8260B</td>
<td>50</td>
<td>mg/kg</td>
<td>0.001</td>
</tr>
</tbody>
</table>
Attachment A – Site-Specific Health and Safety Plan

1.0 Introduction

This Health and Safety Plan includes both site-specific information (including site-specific activities, health hazards, route to hospital, and toxicity information) and information from the Apex’s general Health and Safety Plan.

1.1 Emergency Contact Summary

<table>
<thead>
<tr>
<th>SITE LOCATION</th>
<th>2006 Colt Court, Andover, Kansas</th>
</tr>
</thead>
<tbody>
<tr>
<td>NEAREST HOSPITAL</td>
<td>1124 West 21st Street</td>
</tr>
<tr>
<td></td>
<td>Andover, KS 67002</td>
</tr>
<tr>
<td></td>
<td>(See Figure HSP-1)</td>
</tr>
<tr>
<td></td>
<td>Telephone: (316) 300-4000</td>
</tr>
<tr>
<td>EMERGENCY RESPONDERS</td>
<td>Police Department: 911</td>
</tr>
<tr>
<td></td>
<td>Fire Department: 911</td>
</tr>
<tr>
<td></td>
<td>Ambulance: 911</td>
</tr>
<tr>
<td>EMERGENCY CONTACTS</td>
<td>Apex Companies, LLC: (503)924-4704</td>
</tr>
<tr>
<td></td>
<td>National Response Center: (800)424-8802</td>
</tr>
<tr>
<td></td>
<td>Poison Control Center: (800)222-1222</td>
</tr>
<tr>
<td></td>
<td>Chemtrec: (800)424-9300</td>
</tr>
</tbody>
</table>

In the event of an emergency, call for help as soon as possible. Give the following information:

- WHERE the emergency is - use cross streets or landmarks
- PHONE NUMBER you are calling from
- WHAT HAPPENED - type of injury
- HOW MANY persons need help
- WHAT is being done for the victim(s)
- YOU HANG UP LAST - let the person you called hang up first

2.0 Corporate Health and Safety Plan

The Apex General Health and Safety Plan, together with the included site-specific information, cover each of the 11 required plan elements as specified in Occupational Safety and Health Administration (OSHA) 1910.120, and meet all applicable regulatory requirements. The reader is advised to thoroughly review the entire plan.
3.0 Site Specific Health and Safety Plan

3.1 Site Location and Description

LOCATION: Quail Crossing Neighborhood in Andover, Kansas.

LAND USE OF AREA SURROUNDING FACILITY: Residential

3.2 Site Activity Summary

SITE ACTIVITIES: Hollow-stem auger borehole explorations; sampling of soil.

PROPOSED DATE OF ACTIVITY: April 2014.

POTENTIAL SITE CONTAMINANTS: Benzene, toluene, ethylbenzene, xylenes (BTEX), 2-butanone, gasoline-range organics (GRO), and other volatile organic compounds (VOCs).

POTENTIAL ROUTES OF ENTRY: Skin contact with soil, incidental ingestion of soil, and inhalation of dust and volatiles.

PROTECTIVE MEASURES: Engineering controls, safety glasses, safety boots, hard hat, gloves, protective clothing (including fire-resistant clothing), and respirators, as necessary.

MONITORING EQUIPMENT: Photoionization detector (PID) with 10.2 eV lamp and olfactory indications.

3.3 Chain of Command

The chain of command for Health and Safety in this project involves the following individuals:

CORPORATE H&S MANAGER: Adam Reese

PROJECT MANAGER: Sam Jackson.

PROJECT H&S OFFICER: Sam Jackson

FIELD H&S MANAGER: Michael Whitson
3.4 Hazard Analysis and Applicable Safety Procedures

The following work tasks will be performed:

- Soil borings; and
- Soil sampling.

The hazards associated with the activities listed above are discussed in detail below.

3.4.1 Soil Borings

Drilling and activities will be conducted with appropriate protection, as discussed under personal protective equipment requirements. Employees are cautioned to stand clear of all equipment. Noise protection must also be available and used whenever drilling activities are in progress. In addition, exclusion zones will be established for worker protection.

Underground Utilities. Any underground activity that disturbs soil has the potential for disrupting underground utilities. Immediately stop work and evacuate the area pending further evaluation:

- If gas or vapor venting occurs during the activity;
- If the odor of natural gas is detected; or
- If it is suspected a pipeline or utility service has been damaged.

In addition, contact the proper authorities as necessary, and report the incident to the project manager.

If gas or vapor venting occurs from a soil boring, well installation, excavation, or other source, immediately move upwind of the source. If necessary, use respiratory protection. If the odor of natural gas is detected or if it is suspected that a pipeline has been hit, immediately stop work, evacuate the area, and contact the proper authorities.

Never continue to work in an area, even if PID readings, LEL, and/or hydrogen sulfide tests are acceptable, if you begin to notice strange odors or symptoms of overexposure (such as dizziness, nausea, tearing of the eyes, etc.). Do not resume work until testing shows the hazard has been removed.

Slips, Trips, and Falls. The work area will include uneven surfaces, surfaces with limited traction, and debris may be present. Caution will be used to avoid slips, trips and falls.
3.4.2 Soil Sampling

Soil sampling will occur under the assumption the media is contaminated and appropriate personnel protection is required.

3.4.3 Air Monitoring and Action Levels

Air monitoring will be conducted to determine possible hazardous conditions and to confirm the adequacy of personal protection equipment. The results of the air monitoring will be used as the basis for specifying personal protective equipment and determining the need for upgrading protective measures.

Air monitoring equipment will be calibrated prior to use (where applicable) as specified by the instrument manuals and results will be documented in the instrument log. All equipment will be maintained as specified by the manufacturer or more frequently as required by use conditions. Repair records will be maintained with the instrument log.

PID Monitoring. Air monitoring will be conducted with a PID with 10.2 eV lamp, or equivalent, to measure organic vapor concentrations during site work activities. Background PID measurements will be taken prior to the start of activities to quantify levels associated with the ambient air space in the vicinity of the site. After the completion each open borehole, a separate PID measurement will be collected from within the borehole to quantify the potential for VOCs to be released into the breathing space. If any of these workspace PID measurements are elevated relative to the previously measured background levels, then detector tube readings will be collected from the breathing space (described below). If the detector tube readings exceed the National Institute for Occupational Safety and Health (NIOSH) recommended exposure limit (REL) concentrations then site workers exposed to these levels will use air purifying respirators as appropriate. If detector tubes readings are below the REL concentrations, then a PID measurement will be collected from the breathing space. If PID measurements are elevated in the breathing zone above background concentrations, then site workers exposed to these levels will use air purifying respirators as appropriate. If measured concentrations exceed immediately dangerous to life and health (IDLH) concentrations, site work will cease and personnel will vacate the work area pending re-evaluation of the situation by the Health and Safety Manager.

Detector Tubes. If VOCs are detected as described above, air monitoring within the breathing space will be conducted for specific compounds of concern with detector tubes for each identified compound. Selected compounds have low permissible exposure limits (PEL) or register less effectively with the PID. Specific detector tubes will be collected for benzene.

Olfactory. If olfactory senses detect any unfamiliar odor, work will stop until an assessment can be made to determine whether the need exists to upgrade protective measures.
3.5 Chemicals of Concern

Based on site information gathered to date, the following chemicals may be present at this site:

- GRO;
- BTEX and 2-butanone;
- Other VOCs.

3.5.1 Toxicity Information

Pertinent toxicological properties of these chemicals are discussed below. This information generally covers potential toxic effects which may occur from relatively significant acute and/or chronic exposures, and is not meant to indicate that such effects will occur from the planned site activities. In general, the chemicals which may be encountered at this site are not expected to be present at concentrations which could produce significant exposures. The types of planned work activities should also limit potential exposures at this site. Furthermore, appropriate protective and monitoring equipment will be used as discussed below to further minimize any exposures which might occur.

Standards for occupational exposures to these chemicals are included where available. Site exposures are generally expected to be of short duration and well below the level of any of these exposure limits. These standards are presented below:

PEL Permissible exposure limit (OSHA).
REL Recommended exposure limit (NIOSH).
IDLH Immediately dangerous to life and health (NIOSH)
TWA Time-weighted average exposure limit for any 8-hour work shift of a 40-hour work week.
STEL Short term exposure limit expressed as a 15-minute time-weighted average and not to be exceeded at any time during a work day.
C Ceiling exposure limit not to be exceeded at any time during a work day.

Total Petroleum Hydrocarbons. Total petroleum hydrocarbons (TPH) is a term used to describe a broad family of several hundred chemical compounds that originally come from crude oil. In this sense, TPH is really a mixture of chemicals. They are called hydrocarbons because almost all of them are made entirely from hydrogen and carbon. Crude oils can vary in how much of each chemical they contain, and so can the petroleum products that are made from crude oils. Most products that contain TPH will burn. Some are clear or light-colored liquids that evaporate easily, and others are thick, dark liquids or semi-solids that do
not evaporate. Many of these products have characteristic gasoline, kerosene, or oily odors. Because modern society uses so many petroleum-based products (for example, gasoline, kerosene, fuel oil, mineral oil, and asphalt), contamination of the environment by them is potentially widespread. Contamination caused by petroleum products will contain a variety of these hydrocarbons. Because there are so many, it is not usually practical to measure each one individually. However, it is useful to measure the total amount of all hydrocarbons found together in a particular sample of soil, water, or air.

TPH can enter and leave your body when you breathe it in air; swallow it in water, food, or soil; or touch it. Most components of TPH will enter your bloodstream rapidly when you breathe them as a vapor or mist or when you swallow them. Some TPH compounds are widely distributed by the blood throughout your body and quickly break down into less harmful chemicals. Others may break down into more harmful chemicals. Other TPH compounds are slowly distributed by the blood to other parts of the body and do not readily break down. When you touch TPH compounds, they are absorbed more slowly and to a lesser extent than when you breathe or swallow them. Most TPH compounds leave your body through urine or when you exhale air containing the compounds.

The compounds in different TPH fractions affect the body in different ways. Some of the TPH compounds, particularly the smaller compounds such as benzene, toluene, and xylene (which are present in gasoline), can affect the human central nervous system. If exposures are high enough, death can occur. Breathing toluene at concentrations greater than 100 parts per million (ppm) for more than several hours can cause fatigue, headache, nausea, and drowsiness. When exposure is stopped, the symptoms will go away. However, if someone is exposed for a long time, permanent damage to the central nervous system can occur. One TPH compound (n-hexane) can affect the central nervous system in a different way, causing a nerve disorder called “peripheral neuropathy” characterized by numbness in the feet and legs and, in severe cases, paralysis. This has occurred in workers exposed to 500–2,500 ppm of n-hexane in the air. Swallowing some petroleum products such as gasoline and kerosene causes irritation of the throat and stomach, central nervous system depression, difficulty breathing, and pneumonia from breathing liquid into the lungs. The compounds in some TPH fractions can also affect the blood, immune system, liver, spleen, kidneys, developing fetus, and lungs. Certain TPH compounds can be irritating to the skin and eyes. Other TPH compounds, such as some mineral oils, are not very toxic and are used in foods. One TPH compound (benzene) has been shown to cause cancer (leukemia) in people. The International Agency for Research on Cancer (IARC) has determined that benzene is carcinogenic to humans (Group 1 classification). Some other TPH compounds or petroleum products, such as benzo(a)pyrene and gasoline, are considered to be probably and possibly carcinogenic to humans (IARC Groups 2A and 2B, respectively) based on cancer studies in people and animals. Most of the other TPH compounds and products are considered not classifiable (Group 3) by IARC.
Although there are no federal regulations or guidelines for TPH in general, the government has developed regulations and guidelines for some of the TPH fractions and compounds. These are designed to protect the public from the possible harmful health effects of these chemicals. To protect workers, the OSHA has set a legal limit of 500 ppm in the workplace.

EPA regulates certain TPH fractions, products, or wastes containing TPH, as well as some individual TPH compounds. For example, there are regulations for TPH as oil; these regulations address oil pollution prevention and spill response, stormwater discharge, and underground injection control. EPA lists certain wastes containing TPH as hazardous. EPA also requires that the National Response Center be notified following a discharge or spill into the environment of 10 pounds or more of hazardous wastes containing benzene, a component in some TPH mixtures.

Nearly all states have cleanup standards for TPH or components of TPH (common cleanup standards are for gasoline, diesel fuel, and waste oil). Analytical methods are specified, many of which are considered to be TPH methods.

Benzene. Benzene, also known as benzol, is a colorless liquid with a sweet odor. Benzene evaporates into air very quickly and dissolves slightly in water. Benzene is highly flammable. Most people can begin to smell benzene in air at 1.5–4.7 ppm and smell benzene in water at 2 ppm. Most people can begin to taste benzene in water at 0.5–4.5 ppm. Benzene is found in air, water, and soil.

Benzene found in the environment is from both human activities and natural processes. Benzene was first discovered and isolated from coal tar in the 1,800s. Today, benzene is made mostly from petroleum sources. Because of its wide use, benzene ranks in the top 20 in production volume for chemicals produced in the United States. Various industries use benzene to make other chemicals, such as styrene (for Styrofoam® and other plastics), cumene (for various resins), and cyclohexane (for nylon and synthetic fibers). Benzene is also used for the manufacturing of some types of rubbers, lubricants, dyes, detergents, drugs, and pesticides. Natural sources of benzene, which include volcanoes and forest fires, also contribute to the presence of benzene in the environment. Benzene is also a natural part of crude oil and gasoline and cigarette smoke.

Most people are exposed to a small amount of benzene on a daily basis. You can be exposed to benzene in the outdoor environment, in the workplace, and in the home. Exposure of the general population to benzene is mainly through breathing air that contains benzene. The major sources of benzene exposure are tobacco smoke, automobile service stations, exhaust from motor vehicles, and industrial emissions. Vapors (or gases) from products that contain benzene, such as glues, paints, furniture wax, and detergents, can also be a source of exposure. Auto exhaust and industrial emissions account for about 20% of the total nationwide exposure to benzene. About 50% of the entire nationwide exposure to benzene results from...
smoking tobacco or from exposure to tobacco smoke. The average smoker (32 cigarettes per day) takes in about 1.8 milligrams (mg) of benzene per day. This is about 10 times the average daily intake of nonsmokers.

Measured levels of benzene in outdoor air have ranged from 0.02 to 34 parts of benzene per billion parts of air (ppb; 1 ppb is 1,000 times less than 1 ppm). People living in cities or industrial areas are generally exposed to higher levels of benzene in air than those living in rural areas. Benzene levels in the home are usually higher than outdoor levels. People living around hazardous waste sites, petroleum refining operations, petrochemical manufacturing sites, or gas stations may be exposed to higher levels of benzene in air.

Benzene can enter your body through your lungs when you breathe contaminated air. It can also enter through your stomach and intestines when you eat food or drink water that contains benzene. Benzene can enter your body through skin contact with benzene-containing products such as gasoline.

When you are exposed to high levels of benzene in air, about half of the benzene you breathe in leaves your body when you breathe out. The other half passes through the lining of your lungs and enters your bloodstream. Animal studies show that benzene taken in by eating or drinking contaminated foods behaves similarly in the body to benzene that enters through the lungs. A small amount will enter your body by passing through your skin and into your bloodstream during skin contact with benzene or benzene-containing products. Once in the bloodstream, benzene travels throughout your body and can be temporarily stored in the bone marrow and fat. Benzene is converted to products, called metabolites, in the liver and bone marrow. Some of the harmful effects of benzene exposure are believed to be caused by these metabolites. Most of the metabolites of benzene leave the body in the urine within 48 hours after exposure.

After exposure to benzene, several factors determine whether harmful health effects will occur, and if they do, what the type and severity of these health effects might be. These factors include the amount of benzene to which you are exposed and the length of time of the exposure. Most data involving effects of long-term exposure to benzene are from studies of workers employed in industries that make or use benzene. These workers were exposed to levels of benzene in air far greater than the levels normally encountered by the general population. Current levels of benzene in workplace air are much lower than in the past. Because of this reduction, and the availability of protective equipment such as respirators, fewer workers have symptoms of benzene poisoning.

Brief exposure (5–10 minutes) to very high levels of benzene in air (10,000–20,000 ppm) can result in death. Lower levels (700–3,000 ppm) can cause drowsiness, dizziness, rapid heart rate, headaches,
tremors, confusion, and unconsciousness. In most cases, people will stop feeling these effects when they stop being exposed and begin to breathe fresh air.

Eating foods or drinking liquids containing high levels of benzene can cause vomiting, irritation of the stomach, dizziness, sleepiness, convulsions, rapid heart rate, coma, and death. The health effects that may result from eating foods or drinking liquids containing lower levels of benzene are not known. If you spill benzene on your skin, it may cause redness and sores. Benzene in your eyes may cause general irritation and damage to your cornea.

Benzene causes problems in the blood. People who breathe benzene for long periods may experience harmful effects in the tissues that form blood cells, especially the bone marrow. These effects can disrupt normal blood production and cause a decrease in important blood components. A decrease in red blood cells can lead to anemia. Reduction in other components in the blood can cause excessive bleeding. Blood production may return to normal after exposure to benzene stops. Excessive exposure to benzene can be harmful to the immune system, increasing the chance for infection and perhaps lowering the body's defense against cancer.

Benzene can cause cancer of the blood-forming organs. The Department of Health and Human Services (DHHS) has determined that benzene is a known carcinogen. IARC has determined that benzene is carcinogenic to humans, and the EPA has determined that benzene is a human carcinogen. Long-term exposure to relatively high levels of benzene in the air can cause cancer of the blood-forming organs. This condition is called leukemia. Exposure to benzene has been associated with development of a particular type of leukemia called acute myeloid leukemia (AML).

Exposure to benzene may be harmful to the reproductive organs. Some women workers who breathed high levels of benzene for many months had irregular menstrual periods. When examined, these women showed a decrease in the size of their ovaries. However, exact exposure levels were unknown, and the studies of these women did not prove that benzene caused these effects. It is not known what effects exposure to benzene might have on the developing fetus in pregnant women or on fertility in men. Studies with pregnant animals show that breathing benzene has harmful effects on the developing fetus. These effects include low birth weight, delayed bone formation, and bone marrow damage.

The health effects that might occur in humans following long-term exposure to food and water contaminated with benzene are not known. In animals, exposure to food or water contaminated with benzene can damage the blood and the immune system and can even cause cancer.

EPA has set the maximum permissible level of benzene in drinking water at 5 ppb. Because benzene can cause leukemia, EPA has set a goal of 0 ppb for benzene in drinking water and in water such as rivers and
Attachment A – Site-Specific Health and Safety Plan

Lakes. EPA estimates that 10 ppb benzene in drinking water that is consumed regularly or exposure to 0.4 ppb benzene in air over a lifetime could cause a risk of one additional cancer case for every 100,000 exposed persons. EPA recommends a maximum permissible level of benzene in water of 200 ppb for short-term exposures (10 days) for children.

EPA requires that the National Response Center be notified following a discharge or spill into the environment of 10 pounds or more of benzene.

OSHA regulates levels of benzene in the workplace. The maximum allowable amount of benzene in workroom air during an 8-hour workday, 40-hour workweek is 1 ppm. Since benzene can cause cancer, the NIOSH) recommends that all workers likely to be exposed to benzene wear special breathing equipment.

Toluene. Toluene is a clear, colorless liquid with a distinctive smell. It is added to gasoline along with benzene and tolueneylene. Toluene occurs naturally in crude oil and in the tolu tree. It is produced in the process of making gasoline and other fuels from crude oil, in making coke from coal, and as a by-product in the manufacture of styrene. Toluene is used in making paints, paint thinners, fingernail polish, lacquers, adhesives, and rubber and in some printing and leather tanning processes. It is disposed of at hazardous waste sites as used solvent (a substance that can dissolve other substances) or at landfills where it is present in discarded paints, paint thinners, and fingernail polish. You can begin to smell toluene in the air at a concentration of 8 ppm, and taste it in your water at a concentration of 0.04–1 ppm. (One ppm is equivalent to 1 minute in 2 years.)

Toluene can enter your body when you breathe its vapors or eat or drink contaminated food or water. When you work with toluene-containing paints or paint thinners, the toluene can also pass through your skin into your bloodstream. You are exposed to toluene when you breathe air containing toluene. When this occurs the toluene is taken directly into your blood from your lungs. Where you live, work, and travel and what you eat affect your daily exposure to toluene. Factors such as your age, sex, body composition, and health status affect what happens to toluene once it is in your body. After being taken into your body, more than 75% of the toluene is removed within 12 hours. It may leave your body unchanged in the air you breathe out or in your urine after some of it has been chemically changed to make it more water soluble. Generally, your body turns toluene into less harmful chemicals such as hippuric acid.

A serious health concern is that toluene may have an effect on your brain. Toluene can cause headaches, confusion, and memory loss. Whether or not toluene does this to you depends on the amount you take in and how long you are exposed. Low-to-moderate, day-after-day exposure in your workplace can cause tiredness, confusion, weakness, drunken-type actions, memory loss, nausea, and loss of appetite. These symptoms usually disappear when exposure is stopped. Researchers do not know if the low levels of
Toluene you breathe at work will cause any permanent effects on your brain or body after many years. You may experience some hearing loss after long-term daily exposure to toluene in the workplace.

If you are exposed to a large amount of toluene in a short time because you deliberately sniff paint or glue, you will first feel light-headed. If exposure continues, you can become dizzy, sleepy, or unconscious. You might even die. Toluene causes death by interfering with the way you breathe and the way your heart beats. When exposure is stopped, the sleepiness and dizziness will go away and you will feel normal again.

If you choose to repeatedly breathe in toluene from glue or paint thinners, you may permanently damage your brain. You may also experience problems with your speech, vision, or hearing, have loss of muscle control, loss of memory, poor balance, and decreased mental ability. Some of these changes may be permanent.

Toluene may change the way your kidneys work, but in most cases, the kidneys will return to normal after exposure stops. If you drink alcohol and are exposed to toluene, the combination can affect your liver more than either compound alone. This phenomenon is called synergism. Combinations of toluene and some common medicines like aspirin and acetaminophen may increase the effects of toluene on your hearing. In animals, the main effect of toluene is on the nervous system. Animals exposed to moderate or high levels of toluene may also show slightly adverse effects in their liver, kidneys, and lungs.

Several studies have shown that unborn animals were harmed when high levels of toluene were breathed in by their mothers. When the mothers were fed high levels of toluene, the unborn animals did not show any structural birth defects, although some effects on behavior were noted. We do not know if toluene would harm your unborn child if you drink water or breathe air containing low levels of toluene, because studies in people are not comprehensive enough to measure this effect. However, if you deliberately breathe in large amounts of toluene during your pregnancy, your baby can have neurological problems and retarded growth and development.

Studies in workers and in animals exposed to toluene indicate that toluene does not cause cancer. IARC and DHHS have not classified toluene for carcinogenic effects. The EPA has determined that toluene is not classifiable as to its human carcinogenicity.

The federal government has developed regulatory standards and guidelines to protect you from the possible health effects of toluene in the environment. OSHA has set a limit of 100 ppm of toluene for air in the workplace, averaged for an 8-hour exposure per day over a 40-hour work week. The American Conference of Governmental Industrial Hygienists (ACGIH) and NIOSH have recommended that toluene in workplace air not exceed 100 ppm (as an average level over 8 hours).
EPA recommends that drinking water should not contain more than 20 ppm for 1 day, 3 ppm for 10 days, or 1 ppm for lifetime consumption. Any release of more than 1,000 pounds of this chemical to the environment must be reported to the National Response Center.

**Ethylbenzene.** Ethylbenzene is a colorless liquid that smells like gasoline. You can smell ethylbenzene in the air at concentrations as low as 2 ppm. It evaporates at room temperature and burns easily. Ethylbenzene occurs naturally in coal tar and petroleum. It is also found in many products, including paints, inks, and insecticides. Gasoline contains about 2 percent (by weight) ethylbenzene. Ethylbenzene is used primarily in the production of styrene. It is also used as a solvent, a component of asphalt and naphtha, and in fuels. In the chemical industry, it is used in the manufacture of acetophenone, cellulose acetate, diethylbenzene, ethyl anthraquinone, ethylbenzene sulfonic acids, propylene oxide, and -methylbenzyl alcohol. Consumer products containing ethylbenzene include pesticides, carpet glues, varnishes and paints, and tobacco products. In 1994, approximately 12 billion pounds of ethylbenzene were produced in the United States. Ethylbenzene is most commonly found as a vapor in the air. This is because ethylbenzene moves easily into the air from water and soil. Once in the air, other chemicals help break down ethylbenzene into chemicals found in smog. This breakdown happens in less than 3 days with the aid of sunlight. In surface water such as rivers and harbors, ethylbenzene breaks down by reacting with other compounds naturally present in the water. In soil, the majority of ethylbenzene is broken down by soil bacteria. Since ethylbenzene binds only moderately to soil, it can also move downward through soil to contaminate groundwater. Near hazardous waste sites, the levels of ethylbenzene in the air, water, and soil could be much higher than in other areas.

When you breathe air containing ethylbenzene vapor, it enters your body rapidly and almost completely through your lungs. Ethylbenzene in food or water can also rapidly and almost completely enter your body through the digestive tract. It may enter through your skin when you come into contact with liquids containing ethylbenzene. Ethylbenzene vapors do not enter through your skin to any large degree. People living in urban areas or in areas near hazardous waste sites may be exposed by breathing air or by drinking water contaminated with ethylbenzene. Once in your body, ethylbenzene is broken down into other chemicals. Most of it leaves in the urine within 2 days. Small amounts can also leave through the lungs and in feces. Liquid ethylbenzene that enters through your skin is also broken down. Ethylbenzene in high levels is broken down slower in your body than low levels of ethylbenzene. Similarly, ethylbenzene mixed with other solvents is also broken down more slowly than ethylbenzene alone. This slower breakdown will increase the time it takes for ethylbenzene to leave your body.

At certain levels, exposure to ethylbenzene can harm your health. People exposed to high levels of ethylbenzene in the air for short periods have complained of eye and throat irritation. Persons exposed to higher levels have shown signs of more severe effects such as decreased movement and dizziness. No studies have reported death in humans following exposure to ethylbenzene alone. However, evidence from
Animal studies suggest that it can cause death at very high concentrations in the air (about 2 million times the usual level in urban air). Whether or not long-term exposure to ethylbenzene affects human health is not known, because little information is available. Short-term exposure of laboratory animals to high concentrations of ethylbenzene in air may cause liver and kidney damage, nervous system changes, and blood changes. The link between these health effects and exposure to ethylbenzene is not clear because of conflicting results and weaknesses in many of the studies. Also, there is no clear evidence that the ability to get pregnant is affected by breathing air or drinking water containing ethylbenzene, or coming into direct contact with ethylbenzene through the skin. Two long-term studies in animals suggest that ethylbenzene may cause tumors. One study had many weaknesses, and no conclusions could be drawn about possible cancer effects in humans. The other, a recently completed study, was more convincing, and provided clear evidence that ethylbenzene causes cancer in one species after exposure in the air to concentrations greater than 740 ppm that were approximately 1 million times the levels found in urban air. At present, the federal government has not identified ethylbenzene as a chemical that may cause cancer in humans. However, this may change after consideration of the new data.

There are no reliable data on the effects in humans after eating or drinking ethylbenzene or following direct exposure to the skin. For this reason, levels of exposure that may affect your health after eating, drinking, or getting ethylbenzene on your skin are estimated from animal studies. There are only two reports of eye or skin exposure to ethylbenzene. In these studies, liquid ethylbenzene caused eye damage and skin irritation in rabbits. More animal studies are available that describe the effects of breathing air or drinking water containing ethylbenzene.

The federal government develops regulations and recommendations to protect public health. Regulations can be enforced by law. Federal agencies that develop regulations for toxic substances include the EPA, OSHA, and the Food and Drug Administration (FDA). Recommendations provide valuable guidelines to protect public health but cannot be enforced by law. Federal organizations that develop recommendations for toxic substances include the Agency for Toxic Substances and Disease Registry (ATSDR) and NIOSH.

Recommendations and regulations can be expressed in not-to-exceed levels in air, water, soil, or food that are usually based on levels that affect animals; then they are adjusted to help protect people. Sometimes these not-to-exceed levels differ among federal organizations because of different exposure times (an 8-hour workday or a 24-hour day), the use of different animal studies, or other factors.

Recommendations and regulations are also periodically updated as more information becomes available. For the most current information, check with the federal agency or organization that provides it. Some regulations and recommendations for ethylbenzene include the following:
The federal government has developed regulatory standards and guidelines to protect you from possible health effects of ethylbenzene in the environment. EPA's Office of Drinking Water (ODW) set 700 ppb (this equals 0.7 milligrams ethylbenzene per liter of water or mg/L) as the acceptable exposure concentration of ethylbenzene in drinking water for an average weight adult. This value is for lifetime exposure and is set at a level that is expected not to increase the chance of having (noncancer) adverse health effects. The same EPA office (ODW) set higher acceptable levels of ethylbenzene in water for shorter periods (20 ppm or 20 mg/L for 1 day, 3 ppm or 3 mg/L for 10 days). EPA has determined that exposures at or below these levels are acceptable for small children. If you eat fish and drink water from a body of water, the water should contain no more than 1.4 mg ethylbenzene per liter.

EPA requires that a release of 1,000 pounds or more of ethylbenzene be reported to the federal government's National Response Center in Washington, D.C.

OSHA set a legal limit of 100 ppm ethylbenzene in air. This is for exposure at work for 8 hours per day.

NIOSH also recommends an exposure limit for ethylbenzene of 100 ppm. This is for exposure to ethylbenzene in air at work for up to 10 hours per day in a 40-hour work week. NIOSH also set a limit of 125 ppm for a 15-minute period.

**Xylenes.** There are three forms of xylene in which the methyl groups vary on the benzene ring: meta-xylene, ortho-xylene, and para-xylene (m-, o-, and p-xylene). These different forms are referred to as isomers. The term total xylenes refers to all three isomers of xylene (m-, o-, and p-xylene). Mixed xylene is a mixture of the three isomers and usually also contains 6–15% ethylbenzene. Xylene is also known as xylol or dimethylbenzene. Xylene is primarily a synthetic chemical. Chemical industries produce xylene from petroleum. Xylene also occurs naturally in petroleum and coal tar and is formed during forest fires. It is a colorless, flammable liquid with a sweet odor.

Xylene is one of the top 30 chemicals produced in the United States in terms of volume. It is used as a solvent (a liquid that can dissolve other substances) in the printing, rubber, and leather industries. Along with other solvents, xylene is also used as a cleaning agent, a thinner for paint, and in varnishes. It is found in small amounts in airplane fuel and gasoline. Xylene is used as a material in the chemical, plastics, and synthetic fiber industries and as an ingredient in the coating of fabrics and papers. Isomers of xylene are used in the manufacture of certain polymers (chemical compounds), such as plastics.

Xylene evaporates and burns easily. Xylene does not mix well with water; however, it does mix with alcohol and many other chemicals. Most people begin to smell xylene in air at 0.08–3.7 ppm and begin to taste it in water at 0.53–1.8 ppm.
Xylene is most likely to enter your body when you breathe xylene vapors. Less often, xylene enters the body through the skin following direct contact. It is rapidly absorbed by your lungs after you breathe air containing it. Exposure to xylene may also take place if you eat or drink xylene-contaminated food or water. The amount of xylene retained ranges from 50% to 75% of the amount of xylene that you inhale. Physical exercise increases the amount of xylene absorbed by the lungs. Absorption of xylene after eating food or drinking water containing it is both rapid and complete. Absorption of xylene through the skin also occurs rapidly following direct contact with xylene. Absorption of xylene vapor through the skin is lower than absorption of xylene vapor by the lungs. However, it is not known how much of the xylene is absorbed through the skin. At hazardous waste sites, breathing xylene vapors, drinking well water contaminated with xylene, and direct contact of the skin with xylene are the most likely ways you can be exposed. Xylene passes into the blood soon after entering the body.

In people and laboratory animals, xylene is broken down into other chemicals especially in the liver. This process changes most of the xylene that is breathed in or swallowed into a different form. Once xylene breaks down, the breakdown products rapidly leave the body, mainly in urine, but some unchanged xylene also leaves in the breath from the lungs. One of the breakdown products of xylene, methylbenzaldehyde, is harmful to the lungs of some animals. This chemical has not been found in people exposed to xylene. Small amounts of breakdown products of xylene have appeared in the urine of people as soon as 2 hours after breathing air containing xylene. Usually, most of the xylene that is taken in leaves the body within 18 hours after exposure ends. Storage of xylene in fat or muscle may prolong the time needed for xylene to leave the body.

Short-term exposure of people to high levels of xylene can cause irritation of the skin, eyes, nose, and throat; difficulty in breathing; impaired function of the lungs; delayed response to a visual stimulus; impaired memory; stomach discomfort; and possible changes in the liver and kidneys. Both short- and long-term exposure to high concentrations of xylene can also cause a number of effects on the nervous system, such as headaches, lack of muscle coordination, dizziness, confusion, and changes in one's sense of balance. People exposed to very high levels of xylene for a short period of time have died. Most of the information on long-term exposure to xylene is from studies of workers employed in industries that make or use xylene. Those workers were exposed to levels of xylene in air far greater than the levels normally encountered by the general population. Many of the effects seen after their exposure to xylene could have been caused by exposure to other chemicals that were in the air with xylene.

Results of studies of animals indicate that large amounts of xylene can cause changes in the liver and harmful effects on the kidneys, lungs, heart, and nervous system. Short-term exposure to very high concentrations of xylene causes death in animals, as well as muscular spasms, incoordination, hearing loss, changes in behavior, changes in organ weights, and changes in enzyme activity. Long-term exposure of animals to low concentrations of xylene has not been well studied.
Information from animal studies is not adequate to determine whether or not xylene causes cancer in humans. Both the IARC and EPA have found that there is insufficient information to determine whether or not xylene is carcinogenic and consider xylene not classifiable as to its human carcinogenicity.

Exposure of pregnant women to high levels of xylene may cause harmful effects to the fetus. Studies of unborn animals indicate that high concentrations of xylene may cause increased numbers of deaths, decreased weight, skeletal changes, and delayed skeletal development. In many instances, these same concentrations also cause damage to the mothers. The higher the exposure and the longer the exposure to xylene, the greater the chance of harmful health effects. Lower concentrations of xylene are not so harmful.

EPA estimates that, for an adult of average weight, exposure to 10 mg/L (equal to 10 ppm) of water each day for a lifetime (70 years) is unlikely to result in harmful noncancerous health effects. For a long-term but less than lifetime exposure (about 7 years), 27.3 ppm is estimated to be a level unlikely to result in harmful health effects in an adult.

Exposure to 12 ppm xylene in water for 1 day or to 7.8 ppm of xylene in water for 10 days or longer is unlikely to present a health risk to a small child. EPA has proposed a recommended maximum level of 10 ppm xylene in drinking water.

To protect people from the potential harmful health effects of xylene, EPA regulates xylene in the environment. EPA has set a legally enforceable maximum level of 10 mg/L (equal to 10 ppm) of xylene in water that is delivered to any user of a public water system. OSHA has set an occupational exposure limit of 100 ppm of xylene in air averaged over an 8-hour workday and a 15-minute exposure limit of 150 ppm. These regulations also match recommendations (not legally enforceable) of the American Conference of Governmental Industrial Hygienists. NIOSH has recommended an exposure limit (not legally enforceable) of 100 ppm of xylene averaged over a workday up to 10 hours long in a 40-hour workweek. NIOSH has also recommended that exposure to xylene not exceed 150 ppm for longer than 15 minutes. NIOSH has classified xylene exposures of 10,000 ppm as immediately dangerous to life or health.

EPA and the FDA specify conditions under which xylene may be used as a part of herbicides, pesticides, or articles used in contact with food. The EPA has a chronic drinking water health advisory of 27.3 ppm for an adult and 7.8 ppm for a 10-kilogram child.

EPA regulations require that a spill of 1,000 pounds or more of xylene or used xylene solvents be reported to the Federal Government National Response Center.
Driving Directions:
1: Head north on Colt Court toward W Mountain (157 ft)
2: Turn left onto W Mountain (213 ft)
3: Turn right onto N Quail Crossing Street (433 ft)
4: Turn right onto W 21st Street (0.1 mi)
5: Take the 1st left onto Keystone Pkwy
   Destination will be on the left
6: End at Kansas Medical Center:
   1124 West 21st Street
   Andover, KS  67002
   (316) 300-4000
1. PURPOSE AND SCOPE
This Standard Operating Procedure (SOP) provides instructions for standard field screening. Field screening results are used to aid in the selection of soil samples for chemical analysis. This procedure is applicable during all Ash Creek Associates (ACA) soil sampling operations.

Standard field screening techniques include the use of a photoionization detector (PID) to assess for volatile organic compounds (VOCs), for the presence of separate-phase petroleum hydrocarbons using a sheen test. These methods will not detect all potential contaminants, so selection of screening techniques shall be based on an understanding of the site history. The PID is not compound or concentration-specific, but it can provide a qualitative indication of the presence of VOCs. PID measurements are affected by other field parameters such as temperature and soil moisture. Other field screening methods, such as screening for dense non-aqueous phase liquid (DNAPL) using dye or UV light, are not considered “standard” and will be detailed in the site-specific sampling and analysis plan (SAP).

2. EQUIPMENT AND MATERIALS
The following materials are necessary for this procedure:
- PID with calibration gas (record daily calibration/calibration check in field notes);
- Plastic resealable bags (for PID measurement); and
- Glass jars or stainless steel bowls (for sheen testing).

3. METHODOLOGY
Each soil sample will be field screened for VOCs using a PID and for the presence of separate-phase petroleum hydrocarbons using a sheen test. If the presence of DNAPL is suspected, then screening using dye and UV light may also be completed. For information regarding screening using dye or UV light, refer to the site specific sampling and analysis plan.

PID lamps come in multiple sizes, typically 9.8, 10.6, and 11.7 electron volts (eV). The eV rating for the lamp must be greater than the ionization potential (in eV) of a compound in order for the PID to detect the compound. For petroleum hydrocarbons, a lamp of at least 9.8 eV should be used. For typical chlorinated alkenes (dichloroethene, trichloroethene, tetrachloroethene, or vinyl chloride), a lamp of at least 10.6 eV should be used. The compatibility of the lamp size with the site constituents should be verified prior to the field event and will be detailed in the site-specific SAP.

PID Calibration Procedure: The PID used on-site should be calibrated daily or more frequently if needed. Calibration of the PID should be documented in field notes. Calibrations procedures should be conducted according to the manufacturer’s instructions.

PID Screening Procedure:
- Place a representative portion (approximately one ounce) of freshly exposed, uncompacted soil into a clean resealable plastic bag.
- Seal the bag and break up the soil to expose vapors from the soil matrix.
- Allow the bag to sit to reach ambient temperature. Note: Ambient temperature and weather conditions/humidity should be recorded in field notes. Changes in ambient temperature and weather during the field work should also be recorded, as temperature and humidity can affect PID readings.
- Carefully insert the intake port of the PID into the plastic bag.
- Record the PID measurement in the field notes or boring logs.

Sheen Test Procedure:
- Following the PID screen, place approximately one ounce of freshly exposed, uncompacted soil into a clean glass jar or stainless steel bowl.
- Add enough water to cover the sample.
- Observe the water surface for signs of discoloration/sheen and characterize

<table>
<thead>
<tr>
<th>Sheen Type</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>No Sheen (NS)</td>
<td>No visible sheen on the water surface</td>
</tr>
<tr>
<td>Biogenic Film (BF)</td>
<td>Dull, platy/blocky or foamy film.</td>
</tr>
<tr>
<td>Slight Sheen (SS)</td>
<td>Light sheen with irregular spread, not rapid. May have small spots of color/iridescence. Majority of water surface not covered by sheen.</td>
</tr>
<tr>
<td>Moderate Sheen (MS)</td>
<td>Medium to heavy coverage, some color/iridescence, spread is irregular to flowing. Sheen covering a large portion of water surface.</td>
</tr>
<tr>
<td>Heavy Sheen (HS)</td>
<td>Heavy sheen coverage with color/iridescence, spread is rapid, entire water surface covered with sheen. Separate-phase hydrocarbons may be evident during sheen test.</td>
</tr>
</tbody>
</table>
1. PURPOSE AND SCOPE

This Standard Operating Procedure (SOP) describes the methods used for obtaining soil samples for chemical analysis for volatile organic compounds (VOCs) by EPA Method 5035A. Samples collected using the 5035A protocols are not exposed to the atmosphere after sampling thereby reducing the potential for loss of VOCs during sample transport, handling, and analysis. This procedure assumes the use of the PowerStop Handle sampler with disposable EasyDraw Syringes or Terra Core Samplers. This procedure is applicable during all Ash Creek Associates (ACA) soil sampling activities where the 5035A protocols are employed.

2. EQUIPMENT AND MATERIALS

The following materials are necessary for this procedure:

- Sampling equipment (PowerStop Handle, disposable EasyDraw Syringes, Terra Core Samplers)
- Laboratory-supplied sample containers (pre-weighed 40ml VOA vials including labels, preservative, stir bars, etc. [number and type as specified by the lab], two ounce jars)
  - Vials used from ACA stock must be weighed to confirm loss of reagents is less than 0.02 grams. Record vial tare weight in field notes. Discard vials with dates over 6 months old.
- Field documentation materials
- Decontamination materials
- Personal protective equipment (as required by Health and Safety Plan)

3. METHODOLOGY

The project-specific sampling and analysis plan (SAP) will define the specific requirements for 5035A methodology required for a particular site or by a regulatory agency.

Analytical Requirements

- VOCs must be analyzed within 14 days of collection.
- Field preserved samples (e.g., sodium bisulfate or methanol) must be maintained at 4°C.
- Sample collected without preservative (e.g., reagent water) must be frozen or analyzed within 48 hours.

Collection of Samples

- When using the PowerStop Handle, clip the syringe into the handle in one of the three 5 gram positions. Use the heavy position for dense clay, the light position for dry sandy soil, and the medium position for all others.
- Using the handle, push the sampler into the soil to collect the sample. Continue pushing until the soil column has forced the plunger in the syringe to the stopping point or filled the sampler.
- Wipe all debris from the outside of the sampler. The soil plug should be flush with the mouth of the sampler. Remove any excess soil that extends beyond the mouth of the sampler.
- Extrude the 5 gram sample into vial and cap vial immediately. Hold vial at an angle when extruding to minimize splashing. Gently swirl vial for 10 seconds to break up soil particles (do not shake).
- When capping the vial, be sure to remove any soil or debris from the threads of the vial.
- Repeat process for each additional vial.
- Fill a two ounce container (to capacity) for percent total solids determination.
Additional Considerations

- Methanol contamination can occur from adjacent activities (e.g., exhaust from running equipment or vehicles, hot tar roofing, facility operations, etc). Collection and analysis of methanol field blank (e.g., additional methanol vial left open during period of sampling) is recommended.
- Acidification of carbonaceous soils with sodium bisulfate can cause effervescence and loss of VOCs.
- Certain volatile compounds such as 2-chloroethylvinyl ether may be lost by acidification.
- Acidification of certain soils with sodium bisulfate may cause the formation of acetone through oxidation of soil waxes and humic material (e.g., organic materials such as roots).